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TITLE:

Fluorescent Leakage Assay to Investigate Membrane Destabilization by Cell-Penetrating Peptide

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cell-penetrating peptides, membrane, phospholipids, interaction, fluorescence, leakage

SUMMARY:

The fluorescence leakage assay is a simple method that enables the investigation of peptide/membrane interactions in order to understand their involvement in several biological processes and especially the ability of cell-penetrating peptides to disturb phospholipids bilayers during a direct cellular translocation process.

ABSTRACT:

Cell-penetrating peptides (CPPs) are defined as carriers that are able to cross the plasma membrane and to transfer a cargo into cells. One of the main common features required for this activity resulted from the interactions of CPPs with the plasma membrane (lipids) and more particularly with components of the extracellular matrix of the membrane itself (heparan sulphate). Indeed, independent of the direct translocation or the endocytosis-dependent internalization, lipid bilayers are involved in the internalization process both at the level of the plasma membrane and at the level of intracellular traffic (endosomal vesicles). In this article, we present a detailed protocol describing the different steps of a large unilamellar vesicles (LUVs) formulation, purification, characterization, and application in fluorescence leakage assay in order to detect possible CPP-membrane destabilization/interaction and to address their role in the internalization mechanism. LUVs with a lipid composition reflecting the plasma membrane content are generated in order to encapsulate both a fluorescent dye and a quencher. The addition of peptides in the extravesicular medium and the induction of peptide-membrane interactions on the LUVs might thus induce in a dose-dependent manner a significant increase in fluorescence revealing a leakage. Examples are provided here with the recently developed

tryptophan (W)- and arginine (R)-rich Amphipathic Peptides (WRAPs), which showed a rapid and efficient siRNA delivery in various cell lines. Finally, the nature of these interactions and the affinity for lipids are discussed to understand and to improve the membrane translocation and/or the endosomal escape.

INTRODUCTION:

After their discovery in the nineties, cell-penetrating peptides (CPPs) were developed to promote an efficient cellular delivery of cargoes through the plasma membrane^{1,2}. CPPs are usually short peptides, generally 8 to 30 amino acids, having a wide variety of origins. They were first defined as “direct-translocating” carriers, meaning they were able to cross the plasma membrane and to transfer a cargo into cells independently of any endocytotic pathway neither energy requirement nor receptor involvement. However, further investigations revealed that these first observations mainly came from fluorescence overestimation due to the experimental artefact and/or to fixation protocols using methanol leading to strong CPP adherence to the cell membrane and then to internalization of the CPP–dye complex into cells³. Nowadays, it is widely accepted that CPP uptake takes place by both endocytosis and energy-independent translocation^{4–7} depending on different parameters such as the nature of cargo, the used link between CPP and cargo, the studied cell line, etc.

CPPs can be used as transfection agents according to two strategies, either involving a chemical link (covalent strategy) or electrostatic/hydrophobic interactions (non-covalent strategy) between the CPP and its cargo^{8–11}. Although both strategies have shown their efficiency in the cell transfer of several cargoes, the understanding of the mechanism of internalization by CPPs is still under controversy and the balance between endocytosis pathways or direct penetration is still difficult to measure^{12,13}. Although a set of experimental tools and strategies are available to clearly address the involvement of endocytic processes, the direct translocation seems, however, more difficult to characterize since it implies more discrete interactions with plasma membrane components. Biological membranes are usually composed of numerous components, from phospholipids to membrane proteins, which might vary according to the cellular type and/or the environment (stress conditions, cell division, etc.). This diversity of composition, and consequently the absence of a universal cellular membrane model does not enable studies in a single way. However, to circumvent these limitations step-by-step approaches were developed with artificial membrane or membrane extracts. From small unilamellar vesicles to monolayer approaches, every model was clearly pertinent to answer specific questions^{14,15}. Among them, large unilamellar vesicles (LUV) constitute an appropriate membrane mimicking model to study peptide/membrane interactions as being a key point in the internalization process.

In this context, the following protocol describes the investigation of the effects of peptides and peptide/membrane interactions on LUVs integrity through the monitoring of both an anionic fluorescent dye and its corresponding poly-cationic quencher encapsulated in liposomes. This tool is used to study CPP/membrane interactions in order to understand whether they are able to perform a direct membrane translocation. Although usually applied to compare different membrane-interacting peptides, this LUV fluorescence leakage assay could also be used for investigating both CPPs-cargo conjugates (covalent strategy) and CPP: cargo complexes (non-

covalent strategy).

The present protocol will hence be first exemplified with the recently developed tryptophan (W)- and arginine (R)-rich Amphipathic Peptides (WRAP)¹⁶. WRAP is able to form peptide-based nanoparticles to rapidly and efficiently deliver small interfering RNA (siRNA) in several cell lines¹⁶. The fluorescence leakage properties of WRAP peptide alone or siRNA-loaded WRAP-based nanoparticles were monitored to characterize their mechanism of cellular internalization. We showed that their mechanism of internalization mainly involved direct translocation⁷. In a second example, the WRAP peptide was covalently conjugated to the protein/protein interfering peptide iCAL36 (WRAP-iCAL36)¹⁷ and its ability to destabilize membranes was compared in a fluorescence leakage assay to iCAL36 coupled to Penetratin¹⁸ (Penetratin-iCAL36), another CPP.

Finally, the advantages and limitations of the method will be discussed both from a technological point of view and with respect to biological relevance.

PROTOCOL:

1. Preparation of Large Unilamellar Vesicles (LUVs)

1.1. Prepare LUVs for their use as cell membrane mimics for fluorescence leakage assay.

1.2. Mix with a Hamilton glass syringe phosphatidylcholine (DOPC, 786.11 g/mol), sphingomyelin (SM, 760.22 g/mol) and Cholesterol (Chol, 386.65 g/mol) at the molar ratio 4:4:2. The lipid solution is obtained from a stock solution of each lipid solubilized in a methanol/chloroform (3/1; volume/volume) solvent at 25 mg/mL in a 25 mL glass round-bottom flask. Based on 4 μ mole of DOPC, 4 μ mole of SM, and 2 μ mole of Chol, the lipid solution is obtained from stock solution by mixing 126 μ L, 117 μ L, and 31 μ L, respectively.

CAUTION: Methanol is a toxic and inflammable solvent and chloroform is toxic and carcinogenic. Both should be handled with the appropriate protection under a hood.

1.3. Evaporate methanol/chloroform using a rotary evaporator under vacuum during 45–60 min at 60 °C until formation of a dried lipid film.

1.4. Prepare two stock HEPES buffer solutions. Prepare HEPES buffer 1 by mixing 20 mM HEPES (238.3 g/mol) and 75 mM NaCl (58.44 g/mol) and adjust pH to 7.4. Prepare HEPES buffer 2 by mixing 20 mM HEPES and 145 mM NaCl and adjust pH to 7.4. HEPES buffers can be stored at 4 °C for 1 month.

NOTE: It is recommended to check the osmolarity of the buffers using an osmometer.

1.5. Prepare lipid hydration solution by dissolving membrane impermeable fluorescent dye-quencher couple, 8-aminonaphthalene-1, 3, 6-trisulfonic acid, disodium salt at 12.5 mM (ANTS, 427.33 g/mol) and p-xylene-bispyridinium bromide at 45 mM (DPX, 422.16 g/mol) in HEPES buffer

solution. Mixing ANTS with DPX leads to a yellow-colored solution. To achieve the concentrations of 12.5 mM of ANTS and 45 mM of DPX, dissolve 21.4 mg and 76 mg, respectively in 4 mL of HEPES buffer 1.

NOTE: Lipid hydration solution can be stored for 2 weeks at 4 °C by wrapping the tube with aluminum foil.

1.6. Reconstitute multilamellar vesicles (MLV) by resuspending the dried lipid film with 1 mL of the lipid hydration solution and by vortexing until dissolution of the dried lipid film. Ensure that the solution is completely solubilized as small lipid aggregates will negatively impact the preceding steps. Also, check the wall of the glass round-bottom flask to ensure that there is no remaining lipid film.

NOTE: The solution will appear opalescent and light yellow after the solubilization.

1.7. Subject the vesicles to five freeze/thaw cycles. Perform each cycle by putting the glass round-bottom flask for 30 s in liquid nitrogen for freezing step, then leaving it in a water bath for 2 min for thawing step.

NOTE: The temperature of the bath water should be 5–10°C higher than the melting temperature of the lipids.

1.8. Prepare lipid extruder by inserting two filter supports preliminary humidified with HEPES buffer in each polytetrafluoroethylene (PTFE) extruder piece placed in the metal extruder canister.

1.9. Put a HEPES humidified polycarbonate membrane (0.1 µm pore size, 25 mm diameter) on the top of one filter support.

1.10. Assemble the two metal extruder canisters and screw them.

1.11. Place the assembled extruder in the holder and introduce a 1 mL syringe in the appropriate hole at the extremity of each polytetrafluoroethylene extruder piece. Extrusion corresponds to the passage of the liquid tested from one syringe to the other through the polycarbonate membrane.

1.12. Test the extruder with 1 mL of HEPES buffer loaded in one of the 1 mL syringe to ensure that there are no leaks or problems.

1.13. Replace the 1 mL HEPES buffer with the MLV sample.

1.14. Perform extrusion by passaging the MLV sample from one syringe to the other through the polycarbonate membrane at least 21 times to obtain uniform LUVs.

NOTE: Extrusion should be performed at a temperature higher than the melting temperature of the lipid mixture.

2. Purification of LUVs

2.1. Prepare a column purification to remove non-encapsulated ANTS and DPX excess.

2.2. Introduce cross-linked dextran gel (G-50) resuspended in aqueous medium with 0.01% NaN_3 (65 g/mol) in a liquid chromatography column (Luer Lock, Non-jacketed, 1.0 cm x 20 cm, bed volume 16 mL) up to 1 cm below the top of the colorless part of the column.

2.3. Open the tap and let the liquid flow to settle the cross-linked dextran gel.

2.4. Wash the column by eluting with 20 mL of HEPES buffer 2 and discard the output flow of the column.

2.5. Close the tap once the dead volume of solvent above the column is minimized (<100 μL) but sufficient to avoid any drying of the silica.

2.6. Place the freshly extruded LUVs (yellow) on the column and let them enter into the cross-linked dextran gel.

2.7. Continuously add HEPES buffer 2 to the column to perform the LUV purification.

2.8. Elute approximately 2 mL of HEPES buffer 2 (do not forget to regularly fill the top of the column to avoid drying the silica): the free yellow ANTS and DPX solution migrates slower than the liposomes.

2.9. After 1 mL of eluted volume, start collecting purified LUVs in tubes (1.5 mL).

2.10. Observe the drops of eluent from the column and when they become opalescent, they contain liposomes. Change the tube to recover the LUV-containing fraction.

2.11. Elute until the drops are no longer opalescent (~1 mL). Afterwards, elute another 0.5 mL in a separate fraction and then stop eluting.

NOTE: Standards are now available in a wide range of molecular weights, as kits or individual molecular weights to calibrate the elution volume of the LUVs.

2.12. Wrap the tubes with the LUVs in aluminum foil to avoid bleaching of the fluorescence dye.

2.13. Wash the column with 20 mL HEPES buffer 2.

2.14. The LUVs can then be stored for a week at 4 °C.

NOTE: As LUVs stability might depend on LUVs concentration and composition, as well as on ionic strength, the size of the LUVs should be controlled using a dynamic light scattering (DLS) instrument (see section 4. Characterization of LUV Size and Homogeneity) before each test.

3. Quantifying the concentration of LUVs

3.1. Estimate LUV concentration by a phospholipid quantification kit, which enables the evaluation of choline concentration¹⁹. This assay might be applied when phospholipids with choline containing polar head is substantial (>50% of the LUVs).

3.2. Prepare the color reagent by dissolving 18 mg of chromogen substrate in 3 mL of buffer provided.

3.3. Load a polystyrene cuvette, 10 x 10 x 45 mm, with 3 mL of color reagent.

3.4. Use the pure color reagent as blank condition (Blank). Add 20 µL of LUVs sample (Test) or 20 µL of standard solution of known choline concentration (Standard).

3.5. Mix well and incubate for 5 min at 37 °C all conditions (Blank, Test, and Standard).

3.6. Measure the absorbance (optical density, OD) of the test sample and standard solution with the blank solution as the control at 600 nm with a spectrophotometer.

3.7. Check the OD values which enable to estimate the lipid concentration of the LUVs, C[LUV], in choline equivalent compared to the standard of known concentration.

3.8. Perform the calculation using the following equation:

$$C[\text{LUV}] \text{ (mol / l)} = (\text{OD Sample} / \text{OD Standard}) \times C[\text{Standard}] \text{ (mol / l)}$$

NOTE: The phospholipid quantification kit provided a Choline Chloride (139.6 mg/l) standard solution at 54 mg/dL corresponding to molar concentration of C[Standard] = 3.87 mmol/L. OD Sample and OD Standard are the absorbances measured at 600 nm for the LUV and Choline solutions, respectively.

4. Characterization of LUV size and homogeneity

4.1. Perform a measurement using a DLS instrument in order to determine the LUV size (in nm) and polydispersity index (Pdl).

4.2. Program the appropriated “standard operation procedure” (SOP) by indicating the viscosity of the solvent/buffer and the used cuvette.

265 4.3. Place 500 μ L of the LUV solution in a polystyrene semi-micro cuvette.

267 4.4. Insert the polystyrene semi-micro cuvette in a DLS instrument.

269 4.5. At room temperature, measure the size distribution in terms of mean size (Z-average) of
270 the particle distribution and of homogeneity (polydispersity index, Pdl).

272 4.6. All the results are obtained from two independent measurements performed each in
273 three repetitive cycles.

275 NOTE: Standard values for LUVs will be a mean size of 137 ± 7 nm with a Pdl of 0.149 ± 0.041 .

277 5. Preparing peptide solutions

279 5.1. Prepare a stock solution of the peptide, which should be analyzed for the leakage assay.

281 5.2. Dissolve peptide powder (>95% purity) in pure water (e.g., 1 mg peptide in 500 μ L pure
282 water).

284 NOTE: It is recommended to dilute peptides in pure water and to avoid dimethyl sulfoxide (DMSO)
285 solubilization, which could induce artifacts (e.g., membrane permeabilization²⁰).

287 5.3. Vortex the peptide solution for 5 s.

289 5.4. Sonicate the peptide solution in a water sonication bath for 5 min and then centrifuge for
290 5 min at $12,225 \times g$. Collect the supernatant for concentration determination.

292 5.5. Measure the absorbance at 280 nm of three independent peptide dilutions and then
293 calculate peptide concentration using its molar extinction coefficient ϵ (depending on tryptophan
294 and tyrosine content in the peptide sequence) and Beer-Lambert rule.

296 NOTE: If the peptide contains tryptophan and tyrosine, the molar extinction coefficient ϵ is
297 computed on the basis of Tryptophan $\epsilon = 5,690 \text{ M}^{-1}\text{cm}^{-1}$ and Tyrosine $\epsilon = 1,280 \text{ M}^{-1}\text{cm}^{-1}$. If the
298 peptide sequence contains no tryptophan or tyrosine, other colorimetric assay could be
299 performed to measure the concentration (e.g., BCA or Bradford).

301 5.6. Dilute the peptide solution in pure water to a final solution of 100 μ M and store at 4 °C.

303 NOTE: In pure water, no peptide degradation occurs during the 4 °C storage. However, peptide
304 concentration should be measured every 2 weeks to ensure that no water evaporation occurs.

306 6. Fluorescence leakage assay

308 6.1. Fluorescence leakage assay is measured on a spectrofluorometer at room temperature.

Excitation and emission wavelength are fixed at $E_x = 360 \text{ nm} \pm 3 \text{ nm}$ and $E_m = 530 \text{ nm} \pm 5 \text{ nm}$, respectively.

6.2. Dilute LUVs in 1 mL HEPES buffer 2 to a final concentration of $100 \text{ } \mu\text{M}$. Add a magnetic stirrer to homogenize the solution during experiment.

6.3. Measure the LUVs alone during the first 100 s, between $t = 0 \text{ s}$ and $t = 99 \text{ s}$ in order to access the background fluorescence.

NOTE: LUVs alone could also be measured during the whole experiment (15 min) in order to access background fluorescence and potential leaks.

6.4. Thereafter, measure leakage as an increase in fluorescence intensity upon addition of aliquots of peptide solution for the next 900 s (15 min). This protocol is carried out for each concentration of peptide tested from $0.1 \text{ } \mu\text{M}$ to $2.5 \text{ } \mu\text{M}$.

6.5. Finally, 100% fluorescence was achieved by solubilizing the LUVs by addition of $1 \text{ } \mu\text{L}$ of Triton X-100 (0.1%, v/v), resulting in the completely unquenched probe between $t = 1,000 \text{ s}$ and $t = 1,100 \text{ s}$.

7. Quantification of the leakage

7.1. Suppress values obtained after $t = 1,090 \text{ s}$ in order to keep the same number of points for each tested condition.

7.2. Calculate the minimal fluorescence, F_{\min} , by making the average of 50 points between $t = 0 \text{ s}$ and $t = 49 \text{ s}$ (LUVs alone).

7.3. Calculate the maximal fluorescence, F_{\max} , by making the average of 50 points between $t = 1,041 \text{ s}$ and $t = 1,090 \text{ s}$ (LUVs with Triton X-100).

7.4. Calculate the leakage percentage (%Leak) at each time point ($t = x$), according to the following equation:

$$\% \text{Leak}_{(t=x)} = (F_{(t=x)} - F_{\min}) / (F_{\max} - F_{\min}) \times 100$$

7.5. Calculate the average and standard deviation for values obtained with different LUVs preparation ($n \geq 2$) for the same condition.

7.6. Plot the leakage percentage, $\% \text{Leak}_{(t=x)}$, in function of time (s).

REPRESENTATIVE RESULTS:

The principle of the fluorescence leakage assay is shown in the **Figure 1**. In detail, large unilamellar vesicles (LUVs) encapsulating a fluorescent dye and a quencher (no fluorescence signal) are

353 treated with the biomolecule of interest. Due to the interaction of the peptide with lipid
354 membranes, which could imply membrane permeability, reorganization or even rupture, the
355 fluorescence dye and the quencher are released from the LUVs. Subsequent dilutions in the buffer
356 results in an increased fluorescence signal.

357
358 Although this scheme displays a test with free peptides, the advantage of the system lies in the
359 ability to also test cargo-conjugated peptides, peptide-based nanoparticles or other biopolymers,
360 which are suspected to destabilize lipid membranes. Although a preliminary optimization of the
361 protocol especially with regards to the molecules tested is required, this fluorescence leakage
362 assay might be extended to a huge variety of membrane-interacting components. In the present
363 protocol, we show some results obtained with the CPPs and their complexed (non-covalent
364 strategy) or conjugated (covalent strategy) forms. The following examples imply WRAP alone as
365 well as siRNA-loaded WRAP-based nanoparticles and two different peptide-conjugates (WRAP-
366 iCAL36¹⁷ and Penetratin-iCAL36).

367
368 With regard to non-covalent strategy, fluorescence leakage assay with peptides and with their
369 corresponding siRNA-loaded nanoparticles in the presence of LUVs are displayed in **Figure 2**. The
370 vesicles are composed of a mixture of DOPC/SM/Chol (4:4:2) reflecting the plasma membrane as
371 described in Konate et al.²¹ and are usually used to directly evaluate the possibility of lipid bilayer
372 interaction and/or transduction properties of free WRAP and WRAP-based nanoparticles⁷. In the
373 absence of peptides, no leakage is observed (baseline during the first 100 s). Addition of free
374 WRAP on the LUVs induces a significant increase of fluorescence revealing an important LUV
375 leakage and ANTS release. After 15 min, a leakage of $67.8\% \pm 0.4\%$ compared to the Triton
376 condition (positive control) is obtained at the used concentration ($2.5 \mu\text{M}$) of WRAP peptide. It
377 should be noted here that several different concentrations have also been tested and revealed a
378 dose-dependent fluorescence increase, corresponding to a dose-dependent LUV leakage⁷. In
379 contrast, when WRAP is assembled at the same concentration with siRNA to form peptide-based
380 nanoparticles, the leakage is 1.5-fold weaker ($40.5\% \pm 0.5\%$) compared to the free peptide (**Figure**
381 **2**). Similar leakage values have been reported for the RICK peptide (60%) or the RICK:siRNA
382 nanoparticles (28%)²². The difference in values between free peptide compared to nanoparticles
383 might be explained by the fact that, when engaged in the nanoparticles, a substantial part of the
384 peptide is involved in direct interactions with the siRNA, reducing the peptide availability for
385 interactions with lipids.

386
387 Concerning the covalent strategy, fluorescence leakage assays with CPP-conjugates in the
388 presence of LUVs are shown in **Figure 3**. With the same LUV composition [DOPC/SM/Chol (4:4:2)],
389 two conjugated peptides are applied: WRAP-iCAL36 and Penetratin-iCAL36. As previously noticed,
390 no leakage is observed in the absence of peptides. 15 min after injection of $2.5 \mu\text{M}$ of Penetratin-
391 iCAL36, no significant fluorescence increase is detected ($2.3\% \pm 0.7\%$), whereas an addition of 2.5
392 μM of WRAP-iCAL36 induces a net leakage characterized by very strong fluorescence signal
393 ($85.8\% \pm 11.1\%$)¹⁷. These observations indicate that for some peptides, or conjugates, no
394 fluorescence leakage might occur, suggesting no peptide/membrane interactions or no lipid
395 bilayer disturbance. This is in accordance with previously published results showing that
396 Penetratin as well as Tat were not able to destabilize membranes^{23–25}. It should be highlighted

that the membrane destabilization properties of a CPP could change depending on the coupled cargo²⁶.

Furthermore, although WRAP-iCAL36 caused a strong leakage, additional studies do not reveal specific cellular internalization, indicating that these conjugates remained inside the lipids bilayer of the plasma membrane¹⁷. In contrast to the WRAP nanoparticle, we suppose that the WRAP-cargo conjugate is able to destabilize the LUV membrane by sticking between the lipid chains or by forming pores.

These results indicate that the fluorescence leakage assay might reveal the ability of some CPPs to develop peptide/membrane interactions, which could lead to a more or less pronounced membrane permeability. Moreover, these interactions might occur whatever the strategy of cargoes delivery (nanoparticles versus conjugates). Conversely, this method does not discriminate whether a CPP, which does not induce any fluorescence leakage, might still interact with the bilayer or biological membrane of the lipids. This kind of behavior requires additional approaches such as zeta-potential measurements, FRET between peptide and membrane, or tryptophan fluorescence experiments to mention only a few examples.

FIGURE AND TABLE LEGENDS:

Figure 1: Principle of the fluorescence leakage assay. LUVs were loaded with a fluorescent dye (ANTS) in white and its corresponding quencher (DPX) in grey. In the absence of peptides (in red), no fluorescence signal was observed because ANTS fluorescence was quenched by DPX. Addition of peptides on the LUVs induced membrane permeability and the subsequent release of both ANTS and DPX resulting in a significant increase in ANTS fluorescence (yellow).

Figure 2: Fluorescence leakage assay with free WRAP and WRAP:siRNA nanoparticles. Peptide alone and siRNA-loaded nanoparticles were applied on LUVs at the peptide concentration of 2.5 μ M. WRAP-based nanoparticles were formulated at a peptide:siRNA molar ratio of 20 (R 20) as described in Konate et al.^{7,16}. Black arrows show injections of peptide (nanoparticle) and Triton X-100, respectively.

Figure 3: Fluorescence leakage assay with WRAP-iCAL36 and Penetratin-iCAL36 conjugates. Conjugates were applied on LUVs at the concentration of 2.5 μ M. Black arrows show injections of peptide and Triton X-100, respectively.

DISCUSSION:

The presented fluorescence leakage assay is a simple and fast method to address membrane destabilization by cell-penetrating peptide. Easy to do, it also enables an indirect comparison between different membrane-interacting peptides or other membrane-interacting molecules. Concerning critical steps of the protocol, as this assay provides relative values between the baseline (LUVs alone) and maximal fluorescence release (Triton condition), we usually evaluate the concentration of LUVs using the phospholipid quantification kit, which only estimates choline contribution of the LUVs. However, it is also possible to include a more accurate measurement of

LUVs concentration by determining the total phosphorus content by acid digestion as described by Rouser and colleagues²⁷ or by Bartlett²⁸ or by colorimetric method using ammonium ferrothiocyanate complexing phospholipids²⁹. In our hands, it does not have an impact on the relative quantification of leakage and on the comparison between membrane-interacting peptides.

With regard to the use of LUVs, one should also notice the importance to control the state of the baseline of LUV alone in order to check whether they are usable and if they are already showing leaks (characterized by a continuous increase of fluorescence). In the same way, it is important to measure the maximal fluorescence release after each peptide incubation to ensure that no false-negative results were recorded due to an undesired quenching process (e.g., peptide/dye quenching).

In general, according to DLS characterization, LUVs are quite stable for 1 week and should not display a too high fluorescence background. In this context, purification of ANTS/DPX loaded liposomes by gel filtration is also a key point since enabling formation of LUVs without having residual fluorescence that might contribute to the background or produce fluorescence overestimation. Additionally, it is strongly recommended to calibrate the protocol by including a specific positive control that can always give the same leakage percentage. This constitutes an internal control through the different measurements, which might reinforce characterization of the LUVs for each test and increase statistics.

Although the fluorescence leakage assay might provide a fast comparison in membrane-destabilization by CPPs, it is however limited in the interpretation of peptide-membrane interactions since some peptides are able to interact with lipid bilayer without inducing any membrane disturbance or leakage. For these peptides, it is highly recommended to use additional experiments with specific fluorescence markers enabling FRET³⁰.

It should be also noted that other fluorescence dye/quencher pairs (e.g., Tb³⁺/DPA³¹) could be used. Both the methods have their inconveniences: Terbium (III) is not very soluble in water and cannot be used in the presence of phosphate, whereas the ANTS quenching is not a linear function of DPX concentration. Furthermore, other self-quenching materials such as 70 mM Calcein solution³² or dextran-PTS³³ could be handled depending on the membrane defects provoked by the analyzed peptide or compound.

Finally, the main advantage of this fluorescence leakage assay is the ability to test a multitude of potential membrane interacting molecules as well as different membrane compositions, such as endosomal membrane mimics [e.g., dioleoylphosphatidylcholine (DOPC)/dioleoyl-phosphatidylethanolamine (DOPE)/phosphatidylinositol from soybean (PI)/bis(monooleoylglycerol) phosphate (LBPA) at a molar ratio of 5:2:1:2]³⁴, mitochondrial membrane mimics [e.g., 46.5% DOPC, 28.5% phosphatidylethanolamine (PE), 9% phosphatidylinositol (PI), 9% phosphatidylserine (PS), and 7% cardiolipin (CL)³⁵ or any other desired lipid bilayer composition. However, one will first have to ensure stability of the vesicles (no LUV fusion, aggregation, or precipitation, no lipidic falling out of suspension, no negative membrane curvature, etc.) having

a constant mean size close to 100 nm during experiment and storage.

In addition, the advantage of this method compared to extracted membranes (red blood cells, mitochondria, etc.) is the use of purified well-characterized lipids in the absence of proteins. The control of the lipid bilayer composition (plasma, endosomal, or mitochondrial membrane) also enables the insertion of specific membrane proteins (proton pump). Moreover, the ability to control both LUVs internal and external environments allows a clear interpretation of membrane disturbance/leakage. Compared to black-lipid membrane experiments³⁶, which can also visualize membrane permeabilization events, this leakage protocol provides a more simple screening method of membrane-active peptides.

In conclusion, this simple method favors a rapid identification of strong peptide/membrane interactions leading to membrane destabilization. It may be applied to investigate the mechanism of internalization of CPPs and in a more general approach of “membrane-active peptides” such as fusigenic or antimicrobial peptides.

In general, the characterization of the main route used by CPPs to reach the cytoplasm is very complex and requires several distinct approaches, from biophysics to cellular biology. For example, the investigation of WRAP internalization revealed a balance between different mechanisms to enter the cells (endocytosis vs direct translocation) and the fluorescence leakage assay, associated to other methods, has contributed to support a direct penetration process⁷.

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DISCLOSURES:

The authors have no conflicts of interest.

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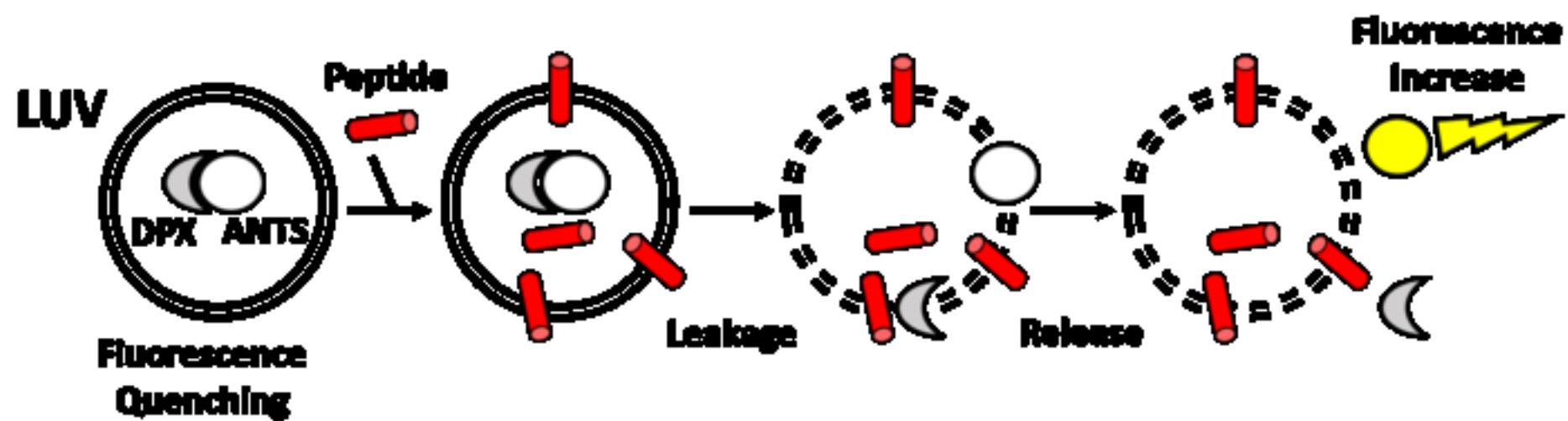


Figure 2

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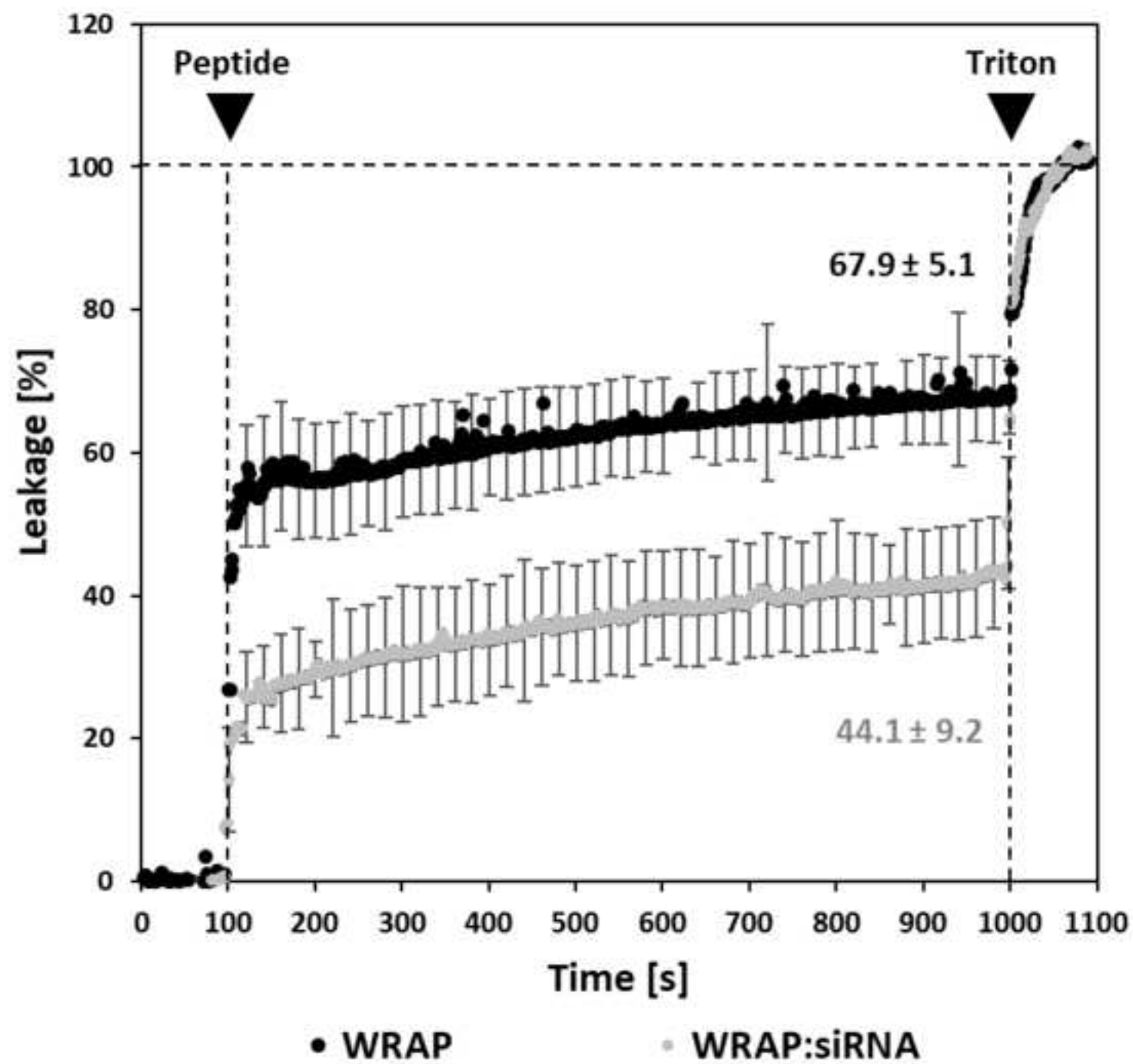
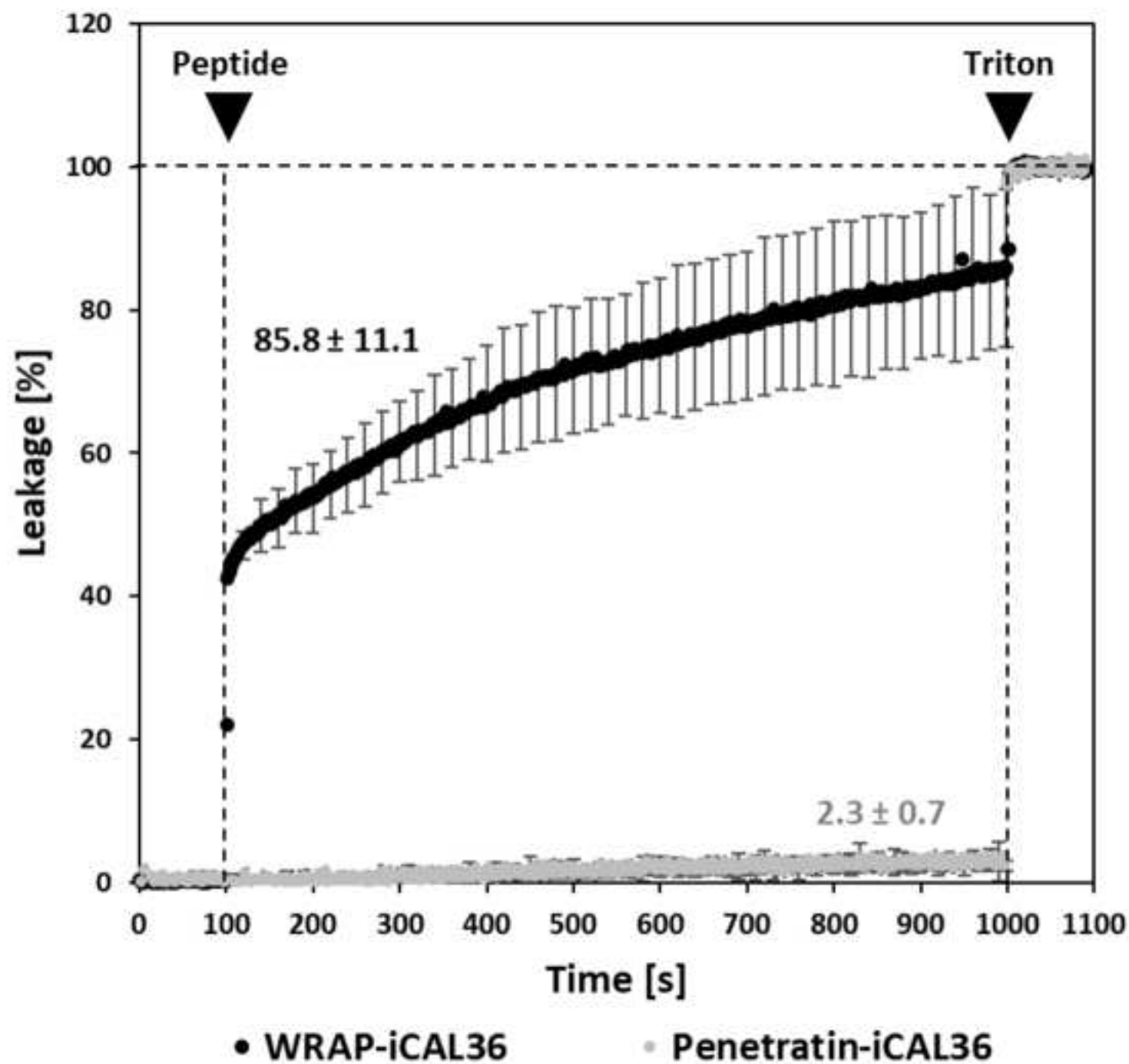


Figure 3



| Name of Material/ Equipment | Company | Catalog Number |
|---|-------------------|----------------|
| 25 mL glass round-bottom flask | Pyrex | |
| 8-aminonaphthalene-1, 3, 6-trisulfonic acid, disodium salt (ANTS) | Invitrogen | A350 |
| Chloroform | Sigma-Aldrich | 288306 |
| Cholesterol | Sigma-Aldrich | C8667 |
| DOPC (dioleoylphosphatidylcholine) | Avanti Polar | 850375P |
| Extruder | Avanti Polar | 610000 |
| Fluorimeter | PTI Serlabo | |
| HEPES | Sigma-Aldrich | H3375 |
| LabAssay Phospholipid | WAKO | 296-63801 |
| liquid chromatography column | Sigma-Aldrich | |
| Methanol | Carlo Erba | 414902 |
| Nuclepore polycarbonate membrane (0.1 µm pore size, 25 mm diameter) | Whatman | 800309 |
| polystyrene cuvette, 10 x 10 x 45 mm | Greiner Bio-One | 614101 |
| polystyrene semi-micro cuvette, DLS | Fisher Scientific | FB55924 |
| p-xylene-bispyridinium bromide (DPX) | Invitrogen | X1525 |
| rotavapor system | Heidolph | Z334898 |
| Sephadex G-50 resin | Amersham | 17-0042-01 |
| Sodium azide (NaN ₃) | Sigma-Aldrich | S2002 |
| Sodium chlorid (NaCl) | Sigma-Aldrich | S5886 |
| Sonicator bath USC300T | VWR | 142-6001 |
| Sphingomyelin | Avanti Polar | 860062P |
| Triton X-100 | Eromedex | 2000-B |
| Zetaziser NanoZS | Malvern | ZEN3500 |

Comments/Description

Protect from light

Protect from air

Protect from light

Protect from air

The authors thank the editorial manager and the reviewers for their comments. The authors answered the questions in a point-by-point rebuttal (see below). Changes in the revised manuscript are highlighted in red.

Editorial comments:

1. The editor has formatted the manuscript to match the journal's style. Please retain and use the attached file for revision.

2. Please address all specific comments marked in the manuscript.

Comments:

We have changed the title in “**Fluorescent leakage assay to investigate membrane destabilization by cell-penetrating peptide**”.

Lines 55-58: We changed the sentence as required.

Lines 116-117: We removed highlight.

Line 139: We are sorry but there is no specific technic to check MLV formation. Only the use of electronic microscopy could give some information but cannot be routinely applied.

Line 155: We added the abbreviation of polytetrafluoroethylene in parentheses (**PTFE**).

Lines 171-174: We have split the point for a better understanding and detailed the extrusion step:

1.13. Replace the 1 mL HEPES buffer with the MLV sample.

1.14. Perform extrusion by passaging the MLV sample from one syringe to the other through the polycarbonate membrane at least 21 times to obtain uniform LUVs.

Line 201: ANTS and DPX are described in point 1.5. In addition we explained the yellow color of the ANTS/DPX mixture both in the point 1.5 and in the point 2.8.

Line 306: We removed the word PTI corresponding to the trademark PTI Serlabo of the fluorimeter.

Line 429: We changed some points in the discussion (see responses to reviewer 1)

3. Please address all reviewers' comments as well.

We have addressed questions/suggestions of reviewer 1 (see below).

4. Once done please ensure that the highlight is no more than 3 pages including headings and spacings.

We have checked the highlighted parts of the manuscript.

Reviewers' comments:

Reviewer #1:

The manuscript has been improved.

Still, I do not agree with the title. Again, the method is useful to study membrane permeability and integrity, not peptide-membrane interaction.

Of course, the presence of leakage implies that the peptide interacted with the membrane, but the opposite is not true: interaction is also possible without leakage, and this was even stated by the authors in their response:

"ii- the CPP could interact at the surface of the LUVs without any membrane perturbation then no leakage"

Authors also state: "Therefore, we are convinced that the described leakage experiment is a suitable assay to indirectly identify peptide-lipid interactions."

Indeed, it is an INDIRECT method: if there is leakage, there is interaction. But, the absence of leakage means nothing. Title is misleading, as well as the description of **the usefulness of the described assays**. This has to be changed.

Comment: We have changed the title of the manuscript and the usefulness of the described assay

Lines 356-358: "Although this scheme displays a test with free peptides, the advantage of the system lies in the ability to also test cargo-conjugated peptides, peptide-based nanoparticles or other biopolymers which are **suspected to destabilize lipid membranes**."

Lines 430-433: "The present fluorescence leakage assay consists in a simple and fast method to address **membrane destabilization by cell-penetrating peptide**. Easy to do, it also enables **an indirect** comparison between different membrane-interacting peptides or other membrane-interacting molecules."

Lines 459-462: "Although the fluorescence leakage assay **might** provide a fast comparison in **membrane-destabilization by CPPs**, it is however limited **in the interpretation of peptide-membrane interactions** since some peptides are able to interact with lipid bilayer without inducing any membrane disturbance or leakage."

LUVs fusion rate depends on LUVs concentration and composition, as well as on ionic strength. Some LUVs suspensions remain stable for long time-periods (in the author hands and in the hands of all persons), but some do not. Since the manuscript points to describing a general method, it is mandatory to include a DLS assay as control before using a stored LUV's suspension.

Comment: We modified the note in point 2.14. with regard to LUVs stability.

Lines 221-223: "**NOTE: As LUVs stability might depend on LUVs concentration and composition, as well as on ionic strength**, the size of the LUVs should be controlled using a dynamic light scattering (DLS) instrument (see section 4. Characterization of LUV Size and Homogeneity) **before each test**."

Reviewer #2:

The authors have responded to the critiques with additional references and changes. From my perspective, it is ready to be accepted. But please see Reviewer 1's comments as he/she had the most number of issues with the premise of the study.

Comment: Please check our comments to reviewer 1.

Reviewer #3:

The authors have properly addressed the concerns that I have raised.