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# DNA Origami–Mediated Substrate Nanopatterning of Inorganic Structures for Sensing Applications --Manuscript Draft--

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1 TITLE: 2 DNA Origami-Mediated Substrate Nanopatterning of Inorganic Structures for Sensing 3 **Applications** 4 5 **AUTHORS AND AFFILIATIONS:** 6 Petteri Piskunen<sup>1</sup>, Boxuan Shen<sup>1</sup>, Sofia Julin<sup>1</sup>, Heini Ijäs<sup>1,2</sup>, J. Jussi Toppari<sup>3</sup>, Mauri A. 7 Kostiainen<sup>1,4</sup>, Veikko Linko<sup>1,4</sup> 8 9 <sup>1</sup> Biohybrid Materials, Biohybrid Materials, Department of Bioproducts and Biosystems, Aalto 10 University, Aalto, Finland 11 <sup>2</sup> University of Jyväskylä, Nanoscience Center, Department of Biological and Environmental 12 Science, University of Jyväskylä, Finland 13 <sup>3</sup> University of Jyväskylä, Nanoscience Center, Department of Physics, University of Jyväskylä, 14 **Finland** 15 <sup>4</sup> HYBER Center of Excellence, Department of Applied Physics, Aalto University, Aalto, Finland 16 17 **Corresponding author:** 18 Veikko Linko (veikko.linko@aalto.fi) 19 20 **Email addresses of co-authors:** 21 Petteri Piskunen (petteri.piskunen@aalto.fi) 22 Boxuan Shen (boxuan.shen@aalto.fi) 23 Sofia Julin (sofia.julin@aalto.fi) 24 Heini Ijäs (heini.ijas@aalto.fi) 25 J. Jussi Toppari (j.jussi.toppari@jyu.fi) 26 Mauri A. Kostiainen (mauri.kostiainen@aalto.fi) 27 28 **KEYWORDS:** 29 DNA nanotechnology, DNA origami, metal nanoparticles, nanolithography, substrate 30 patterning, optics, plasmonics 31 32 **SUMMARY:** 33 Here, we describe a protocol to create discrete and accurate inorganic nanostructures on 34 substrates using DNA origami shapes as guiding templates. The method is demonstrated by 35 creating plasmonic gold bowtie-shaped antennas on a transparent substrate (sapphire). 36 37 ABSTRACT:

Structural DNA nanotechnology provides a viable route for building from the bottom-up using DNA as construction material. The most common DNA nanofabrication technique is called DNA origami, and it allows high-throughput synthesis of accurate and highly versatile structures with nanometer-level precision. Here, it is shown how the spatial information of DNA origami can be transferred to metallic nanostructures by combining the bottom-up DNA origami with the conventionally used top-down lithography approaches. This allows fabrication of billions of tiny nanostructures in one step onto selected substrates. The method is demonstrated using bowtie DNA origami to create metallic bowtie-shaped antenna structures on silicon nitride or sapphire substrates. The method relies on the selective growth of a silicon oxide layer on top of the origami deposition substrate, thus resulting in a patterning mask for following lithographic steps. These nanostructure-equipped surfaces can be further used as molecular sensors (e.g., surface-enhanced Raman spectroscopy (SERS)) and in various other optical applications at the visible wavelength range owing to the small feature sizes (sub-10 nm). The technique can be extended to other materials through methodological modifications; therefore, the resulting optically active surfaces may find use in development of metamaterials and metasurfaces.

#### **INTRODUCTION:**

Structural DNA nanotechnology has rapidly evolved during the recent decade<sup>1,2</sup>, and the most influential development in the field has arguably been the invention of DNA origami<sup>3,4</sup>. The DNA origami technique allows fabrication of virtually any nanoshape with accurate structural features<sup>3,4</sup>. This powerful method can be used in (sub)nanometer-precise spatial arrangement and anchoring of other nano-objects, such as carbon nanotubes<sup>5</sup>, metal nanoparticles<sup>6,7,8,9</sup>, enzymes/proteins<sup>10,11,12,13</sup> and therapeutic materials<sup>14,15,16,17</sup>. Importantly, these structures are not merely static, but they can also be programmed to act in a dynamic manner<sup>18,19</sup>. The countless applications of DNA origami range from drug delivery<sup>20,21,22</sup> to molecular electronics/plasmonics<sup>5,23,24,25</sup> and from materials science<sup>26,27</sup> to novel imaging and calibration techniques<sup>28</sup>.

Besides the applications mentioned above, the extreme spatial resolution of the DNA origami shapes could be harnessed in nanopatterning and delicate nanoscale lithography<sup>29,30</sup>. This protocol describes a lithography method for creating discrete and accurate inorganic nanostructures on substrates using DNA origami templates. These templates can be efficiently produced in various shapes and in large quantities<sup>31</sup>, and deposited effortlessly onto chosen substrates at large scales<sup>32</sup>. These properties allow a highly parallel fabrication of billions of nanostructures in one step as opposed to commonly used but rather slow electron beam lithography or other scanning-based nanofabrication techniques.

Herein, the fabrication process is demonstrated by creating gold bowtie-shaped structures on silicon nitride and sapphire substrates; in other words, the spatial information of DNA origami is transferred to entirely metallic nanostructures. As discussed here, the technique is not limited to the selected bowtie DNA origami structure since the method enables the use of virtually any DNA origami shape. Moreover, with methodical modifications, the technique can be extended to different metals and substrates paving the way towards fabrication of metasurfaces<sup>33</sup>.

The surfaces patterned with the DNA origami-mediated fabrication may serve as versatile sensors; for example, they can be used in surface-enhanced Raman spectroscopy (SERS). As a result of the small dimensions of the individual nanoshapes, the created surfaces may find uses in optical and plasmonic applications at the visible wavelength range.

#### PROTOCOL:

# 1. Design of DNA origami

NOTE: In this protocol, a nanopatterning process is described using a two-dimensional (2D) bowtie DNA origami structure (**Figure 1**)<sup>34</sup>. To design a new DNA origami shape, follow the guidelines below:

1.1. Design the desired shape and the required staple strand sequences of the DNA origami using caDNAno<sup>35</sup>. To produce a flat, single-layer origami, employ the square lattice option of caDNAno and manually adjust the crossover spacing by skipping some bases in the design (see **Figure 1** and the supplemental caDNAno file) to remove the structural twist resulting from the square lattice packing<sup>36,37</sup>.

1.2. Extend the ends of each DNA helix with strands containing poly-T (8 nt) overhangs; this will prevent multimerization of the objects through blunt-end base-stacking interactions (**Figure 1** and supplemental caDNAno file).

1.3. Run a computational analysis of the design. CanDo<sup>38,39</sup> can be used to predict the threedimensional (3D) shape and structural rigidity of the DNA origami. CanDo is also a useful tool to iterate the number of base skips needed for twist correction and to adjust the design accordingly.

1.4. In caDNAno, choose the preferred scaffold length and generate the staple strands
 needed for folding the structure. For the bowtie structure, the 7249 nt long M13mp18 scaffold
 and 205 unique staple strands are used (see the supplemental caDNAno file).

113 114 NOTE: There are also other computational tools available for designing DNA origami 115 structures 40,41,42,43. Depending on the chosen tool/software, other simulation tools may also be 116 used<sup>43,44</sup>. 117 118 2. Assembly of DNA origami 119 120 2.1. Make the stock of staple strands by mixing equal amounts of all the oligonucleotides 121 needed for the bowtie structure (in total 205 staples)<sup>34</sup>. The oligonucleotides should all have 122 the same initial concentration (e.g., 100 µM in RNase free water). 123 124 Prepare the DNA origami folding reaction mixture in 100 µL quantities in a 0.2 mL PCR 2.2. 125 tube by mixing 20 μL of M13mp18 scaffold strand (type p7249, at 100 nM), 40 μL of staple stock solution, where each strand is at 500 nM (which yields ~ 10x molar excess of staples 126 127 compared to the scaffold) and 40 µL of 2.5x folding buffer (FOB). FOB contains Tris - acetic acid 128 - ethylenediaminetetraacetic acid (EDTA) buffer (TAE) supplemented with MgCl<sub>2</sub>. See Table 1 129 for the FOB component concentrations. 130 Anneal the reaction mixture in a thermocycler from 90 °C to 27 °C. Use the thermal 131 2.3. 132 folding ramp presented in Table 2. 133 134 **Purification of DNA origami** 3. 135 136 NOTE: The excess amount of staple strands can be removed from the DNA origami solution 137 using a non-destructive poly(ethylene glycol) (PEG) purification method. The protocol is adapted from Stahl et al.<sup>45</sup>. 138 139 140 3.1. Dilute 200 μL of assembled DNA origami structures with 600 μL of 1x FOB (see Table 1) 141 to obtain a starting volume of 800 µL. 142

3.2. Mix the diluted DNA origami solution 1:1 with 800 μL of PEG precipitation buffer (15%
 PEG 8000 (w/v), 1x TAE, 505 mM NaCl) and mix thoroughly by pipetting back and forth.

146 3.3. Centrifuge the mixture for 30 min at  $14,000 \times g$  and room temperature.

148 3.4. Carefully remove the supernatant using a pipette.

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- 150 3.5. Add 200  $\mu$ L of 1x FOB and mix gently by pipetting. A different amount of 1x FOB can also
- be added to obtain the desired DNA origami concentration.

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3.6. To redissolve the DNA origami structures (small transparent pellet in the bottom of the tube), incubate the PEG purified DNA origami structures overnight at room temperature.

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- 156 3.7. Estimate the DNA origami concentration after PEG purification by measuring the
- absorbance at a wavelength of 260 nm using an UV/Vis spectrophotometer. Use the Beer-
- 158 Lambert law and an extinction coefficient of 1.1·10<sup>8</sup> M<sup>-1</sup> cm<sup>-1</sup> for the calculation<sup>6</sup>. Typical DNA
- origami concentration after PEG purification is 15-20 nM.

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3.8. Store the PEG purified DNA origami structures at 4 °C. The DNA origami structures are usually stable for months so large quantities of stock can be prepared for later use.

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- NOTE: The excess amount of staple strands can also be removed using other purification
- techniques<sup>46</sup>, such as spin-filtration<sup>47</sup>, rate zonal centrifugation<sup>48</sup> and agarose gel extraction<sup>49</sup>.
- 166 The DNA origami structures are stable in a variety of buffer solutions<sup>50</sup>, and if needed, the
- storage medium can be changed after the PEG purification through spin filtration<sup>51</sup>.

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4. Agarose gel electrophoresis

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NOTE: The quality of the folding and the removal of excess staple strands can be verified using agarose gel electrophoresis.

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- 174 4.1. Prepare a ~2% (w/v) agarose gel by adding 1 g of agarose and 45 mL of 1x TAE to an
- 175 Erlenmeyer flask. Heat the mixture in a microwave until the agarose is completely dissolved,
- and a clear solution is produced.

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4.2. Cool down the solution under running water until the flask is comfortable to touch (50–179 60 °C).

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181 4.3. Add 5 mL of 110 mM MgCl<sub>2</sub> and 40  $\mu$ L of ethidium bromide solution (0.58 mg mL<sup>-1</sup>) to 182 the solution and shake the mixture gently.

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184 CAUTION: Ethidium bromide is a potential carcinogen and should be handled with care.

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4.4. Set up the gel casting tray and pour the liquid agarose into the casting tray. Let the gelsolidify at room temperature for at least 30 min.

188 189 4.5. Remove the gel from the casting tray and place it into a gel electrophoresis chamber. Fill 190 the chamber with running buffer (1x TAE with 11 mM MgCl<sub>2</sub>). 191 192 4.6. Add 1 µL of 6x gel loading dye per 5 µL of sample solution and mix thoroughly. Load the 193 samples by carefully pipetting the desired amount of the sample solutions into separate gel 194 pockets. 195 196 4.7. Run the agarose gel at a constant voltage of 95 V for 45 min. Keep the gel 197 electrophoresis chamber on an ice bath for the run to avoid heat damage to the gel.

199 4.8. Visualize the gel under ultraviolet light using a gel imaging system (Figure 2A).

201 5. Substrate preparation (Figure 3A)

NOTE: The following steps are all performed inside a clean room, except for the SiO<sub>2</sub> growth (Step 9). The cleaning steps can also be substituted with a standard piranha-solution based cleaning if this process is not enough to remove all residues from the substrate.

5.1. Cut 7 mm x 7 mm chips from a wafer to be used as a substrate. For SiN, use a silicon saw, a diamond cutter pen or a similar implement. Dicing sapphire ( $Al_2O_3$ ) will require a specialized tool or saw blade. Chip size does not need to be exact.

211 5.2. Cleaning the chips.

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5.2.1. Immerse the diced chips in a glass with hot acetone (acetone heated to 52 °C) and keep
 them heated for at least 15 min. Depending on the starting cleanliness of the substrate, a
 longer time might be necessary.

5.2.2. While they are still in the hot acetone bath, gently rub the chips with a cotton swab to mechanically remove any residue films.

5.2.3. Using tweezers, lift the chips from the hot acetone and use a wash bottle to rinse them with room temperature acetone.

223 5.2.4. Immerse the chips in a glass with isopropanol and ultrasonicate for 2 min.

- 225 5.2.5. Lift the chips out from the isopropanol with tweezers and dry them immediately and 226 thoroughly using a nitrogen flow. Only touch and hold the sides and edges of the chips, as areas 227 covered by the tweezers will not dry properly, leaving potentially residues and other 228 contamination on contact areas. For the best results, use as high flow as possible and hold the 229 chip surfaces parallel to the flow direction. 230 231 5.3. Store the chips in a covered container inside the cleanroom for later use. 232 233 Plasma-enhanced chemical vapor deposition (PECVD) of the amorphous silicon (a-Si) 6. 234 layer (Figure 3B) 235 236 6.1. Place the chips into the PECVD equipment. 237 238 6.2. Set up the deposition parameters to grow roughly 50 nm of amorphous silicon (a-Si). 239 Exact settings vary by equipment model and calibration. See Table 3 for the parameters used 240 here. Run the a-Si deposition program to grow the layer. 241 242 6.3. After processing, store the chips in a covered container in standard clean room 243 conditions. 244 245 7. Oxygen plasma treatment of the a-Si layer (Figure 3B) 246 247 NOTE: This step will make the substrate surface slightly negatively charged and hydrophilic, so 248
- that the DNA origami structures can be later effectively adsorbed to the surface with the help 249 of additional magnesium ions.
  - 7.1. Place the chips into the reactive ion etching (RIE) equipment.
- 253 7.2. Set up the etching parameters to generate oxygen plasma. Again, exact settings vary by 254 equipment model and calibration. See **Table 3** for the parameters used here. Run the oxygen 255 plasma treatment program.
- 257 7.3. Continue to the next step immediately as the effects of the treatment will deteriorate 258 fast. Typically, the substrates should be used within the next 30 min after the plasma 259 treatment.
- 261 8. **Deposition of DNA origami (Figure 3C)**

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- 263 8.1. Prepare a DNA origami mixture for deposition by mixing 5  $\mu$ L of folded/purified DNA origami solution (~20 nM) with 4  $\mu$ L of 1x FOB and 1  $\mu$ L of 1 M MgCl<sub>2</sub>. The resulting solution contains ~10 nM DNA origami and roughly 100 mM of Mg<sup>2+</sup>.
- 266
   267 8.2. Deposit 10 μL of the DNA origami mixture on an oxygen plasma-treated chip and
   268 incubate covered for 5 min at room temperature. Covering prevents unintended drying and
   aids in removing extraneous salt and DNA origami structures later.
- 8.3. After incubation, wash the surface by first pipetting 100 μL of distilled water (e.g.,
   MilliQ) on the chip. Rinse the water back and forth a few times with the pipette, while avoiding
   touching the center of the chip. Remove most of the water from the surface with the pipette.
   This causes only the properly adsorbed origami to remain on the surface.
- 276 8.4. Repeat this washing cycle (steps 8.3) 3 to 4 times.

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- 278 8.5. After washing, dry the sample immediately with a nitrogen flow. Do this the same way as the drying in substrate preparation (step 5). It is important to dry the sample as thoroughly as possible.
- NOTE: The density of deposited structures and thus the density of the metal nanostructures can be modified by adjusting the concentration of DNA origami and Mg<sup>2+</sup> in the deposition solution. Higher Mg<sup>2+</sup> concentration improves DNA origami adhesion and thus increases density, but it will eventually also cause agglomeration of the DNA origami structures. Thus, primarily the DNA origami concentration should be adjusted first.
  - 9. Growth of the SiO<sub>2</sub> mask (Figure 3D)
- NOTE: This step can be performed outside the cleanroom. The following version will yield a negative-tone pattern, but it is possible to modify the process to yield a positive-tone pattern instead. The SiO<sub>2</sub> growth process is adapted from Surwade et al.<sup>52</sup>, developed further by the authors<sup>53</sup>, and finally optimized for this protocol.
- 9.1. Take a sealable desiccator (1.5 L), a Petri dish that fits inside the desiccator (optional)
  and a perforated plate that can function as a platform inside the desiccator.
- 9.2. Take 100 g of silica gel and mix it with 30 g of distilled water in the Petri dish or directly
   in the desiccator. Do this step preferably at least 24 h in advance to allow the silica gel to
   stabilize.

301 302 NOTE: This is used to control the humidity inside the desiccator and therefore also the growth 303 rate and morphology of the SiO<sub>2</sub> film. Higher humidity results in higher rate and coarser 304 structure. Alternatively, the silica gel can be cured in a climatic test chamber. 305 306 9.3. Place the silica gel in the desiccator and separate it with the perforated plate. 307 308 Position the chips with adsorbed DNA origami as well as an open vial of (fresh) 10 mL of 9.4. 309 Tetraethyl orthosilicate (TEOS) and another vial of 10 mL of 25% ammonium hydroxide (NH<sub>4</sub>OH) 310 in the desiccator, on the perforated platform. Set the vials near and on opposite sides of the 311 samples. Preferably use a flask cork or a similar flat pedestal to slightly raise the chips from the 312 platform. 313 314 CAUTION: Both NH<sub>4</sub>OH and TEOS are harmful in case of skin contact and their vapors can cause 315 irritation to both eyes and respiratory organs. Use in a well-ventilated area and wear protective 316 gloves, eye protection and protective clothing. 317 318 9.5. Seal the chamber and incubate for 20 hours at room temperature. This will grow a SiO<sub>2</sub> 319 film on the areas where the DNA origami structures are not located, creating a 10-20 nm 320 patterned mask with DNA origami shaped holes (Figure 4). 321 322 Remove the samples from the chamber after incubation. Store in a covered container. 9.6. 323 Processing can be paused here. Dispose of the used TEOS and NH<sub>4</sub>OH. The batch of silica gel can 324 be used 2-3 times if it is kept sealed inside the desiccator between uses and used within 2-3 325 weeks. 326 327 10. Reactive ion etching (RIE) of SiO<sub>2</sub> and a-Si (Figure 3E) 328 329 Place the chips into the reactive ion etching (RIE) equipment. 330 331 10.2. Set up the etching parameters to only etch 2-5 nm of SiO<sub>2</sub> in order to reveal the a-Si 332 layer beneath the holes in the SiO<sub>2</sub> mask. Exact settings must be determined experimentally for 333 the individual equipment. The parameters used here are presented in Table 3. Run the 334 anisotropic SiO<sub>2</sub> plasma etching program. 335 336 10.3. Set up the etching parameters to pierce through the 50 nm a-Si layer. The parameters

used here are again presented in **Table 3**. Run the isotropic a-Si plasma etching program.

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339	10.4.	Remove samples from RIE equipment and store covered. Processing can be again			
340	o suspended here.				
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342	11.	Physical vapor deposition (PVD) of metals (Figure 3F)			
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344	11.1.	Load the chips into the evaporation chamber of the PVD instrument.			
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346	11.2.	Choose a target metal. First, choose an adhesive metal. Here, 2 nm of chromium (Cr) is			
347	<mark>used.</mark>				
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349	11.3.	Set up the thickness control program for the target material and thickness. The control			
350	metho	d is instrument dependent. Here, a quartz crystal microbalance (QCM) is used. The			
351	measu	red thickness is adjusted by target material density and Z-factor and needs to be			
352	correct	ed by an experimentally determined tooling factor that is specific for the device and			
353	each ta	arget material.			
354					
355	11.4.	Start the electron beam, align the beam to the target and increase beam current until a			
356	<mark>deposi</mark>	tion rate of 0.05 nm/s is reached. Evaporate until a final thickness of 2 nm is reached.			
357					
358	11.5.	Choose a second target metal (e.g. gold) without venting the chamber or interrupting			
359	the pro	ocess. Interruptions or venting will allow the adhesive metal to start oxidizing and			
360	decrea	se its usability as an adhesive.			
361					
362	11.6.	Repeat Steps 11.3 to 11.4. Evaporate until 20 nm is reached. This will create a DNA			
363	<mark>origam</mark>	i shaped metal structure through the $SiO_2$ mask holes with a total height of 22 nm.			
364					
365	11.7.	Vent the chamber and remove samples.			
366					
367	11.8.	Processing can be paused here if the samples are stored covered.			
368					
369	12.	Lift-off with hydrofluoric acid (HF) (Figure 3G)			
370					
371	12.1.	Pour 50% HF-based etchant solution in a suitable plastic container. No HCl should be			
372	used fo	or the mixture, since HCl would etch the Cr in the sample.			
373					
374	CAUTIO	ON: HF is extremely corrosive, causes severe irritation and burns and can be fatal on skin			
375	contact or if inhaled. Use HF only in a dedicated fume hood or ventilated wet bench with a				

376	protective apron, chemical resistant gloves and face visor, or otherwise full chemical				
377	protection.				
378					
379	12.2.	Immerse the samples in the HF-based etchant and stir gently with plastic tweezers.			
380					
381	12.3.	Wait for the SiO <sub>2</sub> layer to etch completely and the metal layer to detach. The time will			
382	vary n	oticeably depending on the density of the mask holes. A higher number of holes will			
383	transla	te to faster etching. If the metal layer is difficult to peel off, brief ultrasonication for 5 to			
384	10 s ca	n be used.			
385					
386	12.4.	Once the metal film detaches, rinse the samples with double-distilled water and			
387	<mark>isopro</mark>	<mark>panol.</mark>			
388					
389	12.5.	After rinsing, dry the samples with a nitrogen flow the same way as instructed for the			
390	<mark>substr</mark>	ate preparation (step 5). Avoid tweezers contact with the chip center, as that may			
391	destro	y the formed nanostructures.			
392					
393	NOTE:	Samples can be stored and processing suspended here.			
394					
395	13.	RIE of remaining a-Si (Figure 3H)			
396					
397	13.1.	Place the chips into the reactive ion etching (RIE) equipment.			
398					
399	13.2.	Set up the etching parameters for thorough removal of all 50 nm of a-Si. The parameters			
400	can be	the same as in Step 10, but a slightly longer etching time (40 s) can be used to ensure			
401	remov	al of all a-Si. See <b>Table 3</b> for the parameters used here. Run the isotropic a-Si plasma			
402	<mark>etchin</mark>	g program to remove remaining a-Si.			
403					
404	13.3.	Remove samples from RIE equipment and store covered. This will conclude sample			
405	<mark>proces</mark>	<mark>sing.</mark>			
406					
407	14.	Atomic force microscopy (AFM)			
408					
409	NOTE:	Atomic force microscopy and scanning electron microscopy can be used to monitor the			
410	succes	s of film growth and patterning as well as to image folded DNA origami structures (Figure			
<b>411</b>	<b>2B C)</b> The following sample preparation step can be skipped if processed samples from Steps				

5-13 are imaged.

415 416 14.1.1. To image the folded DNA origami, take a chip of mica substrate. 417 418 14.1.2. Attach the mica chip to a glass microscope slide using an adhesive. 419 420 14.1.3. Prepare 10 μL of DNA origami solution by diluting the ~20 nM DNA origami stock 50 421 times in 1x FOB to a concentration of approximately 0.4 nM. The dilution is carried out in order 422 to prevent overcrowding the substrate. 423 424 14.1.4. Peel the top layer of the mica sheet off with weak tape to obtain a freshly cleaved, 425 charged surface. 426 427 14.1.5. Deposit the diluted DNA origami solution on the freshly cleaved mica and incubate the 428 sample covered for 1 min at room temperature. 429 430 14.1.6. After incubation, wash the surface 3-4 times with 100 μL of distilled water using a 431 pipette. This causes only the properly adsorbed origami to remain on the surface. 432 433 14.1.7. Deposit 100  $\mu$ L of distilled water on the mica surface. 434 435 14.1.8. Tilt and sharply tap the microscope slide on the table to detach most of the water. 436 437 14.1.9. Repeat this washing cycle 3-4 times. 438 439 Dry the sample thoroughly with a nitrogen flow immediately after washing. The 14.1.10. 440 sample is then ready for AFM imaging. 441 442 14.2. Place the DNA origami samples or the processed chips into an AFM and perform scans. 443 A scan size of 1-10 µm is suitable to properly resolve the structures. 444 445 **15.** Scanning electron microscopy (SEM) 446 447 15.1. Place the samples into a SEM. The processed chips can be used as they are(further 448 sample preparation is not needed).

15.2. Choose the acceleration voltage. Use low voltages (5-10 kV) to reduce charging effects

since the sample substrate ( $Al_2O_3$  or SiN) is an insulator.

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14.1. Sample preparation for AFM

452 453 15.3. Scan any areas of interest. Minimize scanning times to reduce charging and to avoid 454 deposition of contamination. 455 456 **REPRESENTATIVE RESULTS:** 457 A schematic figure of the bowtie DNA origami design and its structural details are shown in 458 Figure 1. Agarose gel electrophoresis and AFM are used to analyze the DNA origami folding and 459 the quality of PEG purification (Figure 2). The process flow of the nanolithography steps is 460 displayed in Figure 3. Representative AFM images after SiO<sub>2</sub> mask growth are shown in Figure 4 461 (this step is depicted in Figure 3D), while SEM images of the final metal nanostructures can be 462 seen in Figure 5 (this step is depicted in Figure 3H). Figure 6 demonstrates the optical 463 functionality of the metallic nanostructures templated by the bowtie DNA origami. 464 465 Table 1: Composition of the folding buffer (FOB). 466 467 Table 2: Thermal ramp for the bowtie origami folding. After annealing, the origami will be 468 stored at 12 °C until the program is manually stopped. 469 470 Table 3: Process parameters for plasma-enhanced chemical vapor deposition (PECVD) and 471 reactive ion etching (RIE). The process parameters for these devices are specific to individual 472 instruments and they may need to be adapted when used. 473 474 Figure 1: Design of the bowtie DNA origami. (A) Schematic representation of the bowtie 475 origami design in which the core structure is shown as double helices and the polyT-overhangs 476 are depicted as wavy lines. (B) Screenshot of a part of the bowtie origami design in the 477 caDNAno software. The red crosses denote the base pair skipping for the twist correction, and 478 the T<sub>8</sub>-overhangs are added to prevent blunt-end base-stacking. 479 480 Figure 2: Characterization of the bowtie DNA origami structure. (A) Agarose gel 481 electrophoresis of the bowtie structure before and after poly(ethylene glycol) (PEG) 482 purification. The 7249 nucleotides long scaffold is used as reference. (B) Atomic force 483 microscopy (AFM) image of the bowtie structures before the purification. (C) AFM image of the 484 bowtie structures after PEG purification. 485 486 Figure 3: Scheme of fabrication process flow (the dimensions are not in scale). (A) Dice and 487 clean the substrate. (B) Deposit an a-Si layer by plasma-enhanced chemical vapor deposition 488 (PECVD). \*It is possible to employ an additional sacrificial layer under the a-Si to enable lift-off

with etchant other than HF. (C) Treat the sample surface with O<sub>2</sub> plasma and deposit DNA

490 origami onto it. (D) Grow the SiO<sub>2</sub> mask in desiccator. (E) Etch a thin layer of SiO<sub>2</sub> and through 491 the a-Si underneath it by reactive ion etching (RIE). (F) Deposit metal through the mask by 492 physical vapor deposition (PVD). (G) Lift-off with HF. (H) remove the remaining a-Si by RIE. 493 494 Figure 4: Representative AFM images of SiO<sub>2</sub> film with the DNA origami shaped pattern. (A) 495 10 µm x 10 µm scanning area demonstrates the high yield of the pattern formation. (B) A closer 496  $3 \mu m \times 3 \mu m$  scan shows the accurate individual patterns in the SiO<sub>2</sub> film. 497 498 Figure 5: Representative scanning electron microscopy (SEM) images of metallic 499 nanostructures templated with structurally different DNA origami. (A) Cross-shaped DNA 500 origami, i.e., so-called Seeman tile origami<sup>54</sup>. (B) Bowtie antennas. (C) Chiral double-L (CDL) structures. Insets show individual structures with box sizes of 150 nm x 150 nm. The fabrication 501 502 yield of exact structures is up to 76% for the bowtie origami and ~ 50% for the other structures 503 displayed here<sup>34</sup>. This figure has been adapted and modified from Shen et al.<sup>34</sup>. The figure is 504 reproduced with permission of the authors and published by The American Association for the 505 Advancement of Science, 2018. 506 507 Figure 6: Representative optical/functional properties of resulting nanostructures. (A) 508 Localized surface plasmon resonance (LSPR) measurements of an individual gold bowtie 509 structure with different polarization (color coded as orange and blue). The solid lines are 510 measured spectra and the dashed lines are simulation results. Insets show the SEM image of 511 the measured particle (left) and the model used for simulation (right). (B) Surface enhanced 512 Raman spectroscopy (SERS) of rhodamine 6G and 2,2-bipyridine measured on a surface covered 513 with bowtie nanostructures. The baseline of each sample shows the signal level when the 514 nanostructures were absent. This figure has been adapted and modified from Shen et al.<sup>34</sup>. The 515 figure is reproduced with permission of the authors and published by The American Association 516 for the Advancement of Science, 2018. 517 518 Supplemental File 1: CaDNAno file 519 520 Supplemental File 2: m13mp18 sequence 521 522 **Supplemental File 3: Staple strand sequence** 523 524 **DISCUSSION:** 525 The protocol provides great freedom and accuracy in the shape of produced nanostructures. By 526 changing the design of the DNA origami, the shape of the metal nanostructures can be

controlled. The final, exact shape of the metal structures is additionally determined by the mask

growth step (Step 9) and to a lesser degree by the mask etching (Step 10) should it not be anisotropic. If the mask growth time is extended enough, the holes in the mask will start to grow shut. This can be used to omit the thinnest features of some structures and control gap sizes, as demonstrated in Shen et al.<sup>34</sup> with separated triangles of the bowtie origami (**Figures 5B**). Conversely, thinner shapes can be better preserved by shortening the oxide growth time. This means that it is possible to tune the optical properties displayed in **Figure 6**, not just by changing the used origami design, but also by tuning the SiO<sub>2</sub> film growth.

If the mask thickness is changed significantly, that change must also be reflected in the  $SiO_2$  RIE step. Only a very thin layer of  $SiO_2$  should be etched (2-5 nm) to barely pierce through the mask holes. This is the most sensitive and crucial part of the whole process. Since the etching time is extremely short, only 10-20 s, exact settings must be experimentally determined when first attempted with new equipment. This is also true for Step 10.4 as some  $SiO_2$  is also etched during the a-Si etching. The extent of etched  $SiO_2$  is determined by the selectivity of the used a-Si etch parameters, equipment and even individual equipment calibrations. Care should be taken not to etch away the entire  $SiO_2$  layer during these two processes.

Another sensitive step is the  $SiO_2$  growth. The growth process is dependent on both the chamber humidity and the current activity of the used TEOS. TEOS degrades as it adsorbs water from the air, causing it to become less effective with age. This can manifest as a significantly slower, less controllable growth rate within months even with proper storage of the chemical.<sup>34</sup> If the resulting  $SiO_2$  layer is thinner than intended, this can indicate a problem with TEOS rather than chamber humidity. While a lower humidity can also result in lower growth rate and thinner film, the resulting film should also be smoother than normal. Meanwhile a coarse grained and rough layer would conversely indicate a problem with high humidity.

It is also possible to perform this protocol on any other freely chosen substrate with two requirements: It must tolerate both HF etching (Step 12) and the 200-300 °C temperatures of PECVD (Step 6). The temperature can be safely lowered to 100 °C for the PECVD of a-Si if a more sensitive substrate is used, but HF cannot be avoided if the protocol is followed exactly as described. To circumvent HF, the application of an additional sacrificial layer would be required. If the requirement of the HF etching is removed, this protocol would become compatible with a wider selection of substrate materials and metals.

As this protocol consists of commonly used and robust micro- and nanofabrication processes, it could be combined with any number of other microfabrication protocols where small feature sizes and complex metal shapes are desired. In the near future, especially with the coming of low-cost DNA origami mass-production<sup>31</sup>, there is potential for this method to facilitate both

566 general use and high-throughput nanopatterning for interface-based nanophotonics and

plasmonics<sup>55</sup>.

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#### DISCLOSURES:

578 The authors have nothing to disclose.

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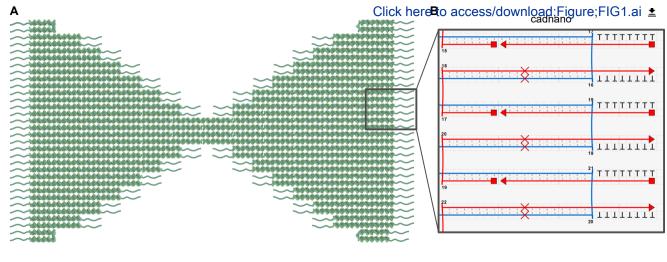
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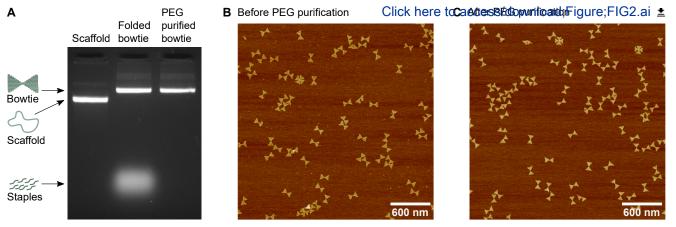
- 581 1. Seeman, N. C., Sleiman, H. F. DNA nanotechnology. *Nature Reviews Materials*. **3** (1),
- 582 17068 (2017).
- 583 2. Linko, V., Dietz, H. The enabled state of DNA nanotechnology. Current Opinion in
- 584 *Biotechnology*. **24** (4), 555-561 (2013).
- 585 3. Rothemund, P. W. K. Folding DNA to create nanoscale shapes and patterns. *Nature*. **440**
- 586 (7082), 297-302 (2006)
- 4. Hong, F., Zhang, F., Liu, Y., Yan, H. DNA Origami: Scaffolds for Creating Higher Order
- 588 Structures. *Chemical Reviews.* **117** (20), 12584-12640 (2017).
- 589 5. Maune, H. T. et al. Self-assembly of carbon nanotubes into two-dimensional geometries
- using DNA origami templates. *Nature Nanotechnology.* **5** (1), 61-66 (2010).
- 591 6. Hung, A. M. et al. Large-area spatially ordered arrays of gold nanoparticles directed by
- 592 lithographically confined DNA origami. *Nature Nanotechnology.* **5** (2), 121-126 (2010).
- 593 7. Kuzyk, A. et al. DNA-based self-assembly of chiral plasmonic nanostructures with
- tailored optical response. *Nature*. **483** (7389), 311-314 (2012).
- 595 8. Zhang, T. et al. 3D DNA Origami Crystals. *Advanced Materials*. **30** (28), 1800273 (2018).
- 596 9. Julin, S. et al. DNA origami directed 3D nanoparticle superlattice via electrostatic
- 597 assembly. *Nanoscale.* **11** (10), 4546-4551 (2019).
- 598 10. Fu, J., Liu, M., Liu, Y., Yan, H. Spatially-Interactive Biomolecular Networks Organized by
- Nucleic Acid Nanostructures. Accounts of Chemical Research. 45 (8), 1215-1226 (2012).
- 600 11. Linko, V. et al. DNA-based enzyme reactors and systems. *Nanomaterials*. **6** (8), 139
- 601 (2015).

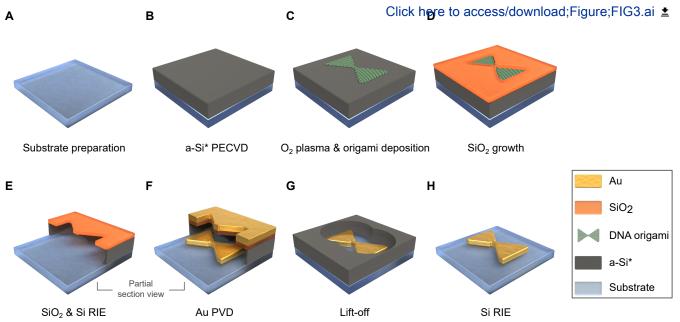
- 602 12. Ramakrishnan, S., Subramaniam, S., Stewart, A. F., Grundmeier, G., Keller, A. Regular
- Nanoscale Protein Patterns via Directed Adsorption through Self-Assembled DNA Origami
- 604 Masks. ACS Applied Materials & Interfaces. **8** (45), 31239-31247 (2016).
- 605 13. Grossi, G., Jaekel, A., Andersen, E. S., Saccà, B. Enzyme-functionalized DNA
- 606 nanostructures as tools for organizing and controlling enzymatic reactions. MRS Bulletin. 42
- 607 (12), 920-924 (2017).
- 608 14. Douglas, S. M., Bachelet, I., Church, G. M. A logic-gated nanorobot for targeted
- 609 transport of molecular payloads. *Science*. **335** (6070), 831-834 (2012).
- 610 15. Li, S. et al. A DNA nanorobot functions as a cancer therapeutic in response to a
- 611 molecular trigger in vivo. Nature Biotechnology. **36** (3), 258-264 (2018).
- 612 16. Zhao, Y.-X. et al. DNA origami delivery system for cancer therapy with tunable release
- 613 properties. ACS Nano. 6 (10), (2014) 8684-8691 (2012).
- 614 17. Kollmann, F. et al. Superstructure-Dependent Loading of DNA Origami Nanostructures
- 615 with a Groove-Binding Drug. ACS Omega. **3** (8), 9441-9448 (2018).
- 616 18. Zhang, D. Y., Seelig, G. Dynamic DNA nanotechnology using strand-displacement
- 617 reactions. *Nature Chemistry*. **3** (2), 103-113 (2011).
- 618 19. Ijäs, H., Nummelin, S., Shen, B., Kostiainen, M. A., Linko, V. Dynamic DNA Origami
- Devices: from Strand-Displacement Reactions to External-Stimuli Responsive Systems.
- 620 International Journal of Molecular Sciences. 19 (7), 2114 (2018).
- 621 20. Li, J., Fan, C., Pei, H., Shi, J., Huang, Q. Smart Drug Delivery Nanocarriers with Self-
- 622 Assembled DNA Nanostructures. Advanced Materials. 25 (32), 4386-4396 (2013).
- 623 21. Linko, V., Ora, A., Kostiainen, M. A. DNA Nanostructures as Smart Drug-Delivery Vehicles
- and Molecular Devices. Trends in Biotechnology. 33 (10), 586-594 (2015).
- 625 22. Jiang, Q., Liu, S., Liu, J., Wang, Z. G., Ding, B. Rationally Designed DNA-Origami
- 626 Nanomaterials for Drug Delivery In Vivo. Advanced Materials. DOI: 10.1002/adma.2018047585
- 627 (2018).
- 628 23. Shen, B., Linko, V., Dietz, H., Toppari, J. J. Dielectrophoretic trapping of multilayer DNA
- 629 origami nanostructures and DNA origami-induced local destruction of silicon dioxide.
- 630 *Electrophoresis.* **36** (2), 255-262 (2015).
- 631 24. Kuzyk, A., Jungmann, R., Acuna, G. P., Liu, N. DNA Origami Route for Nanophotonics. ACS
- 632 *Photonics*. **5** (4), 1151-1163 (2018).
- 633 25. Liu, N., Liedl, T. DNA-Assembled Advanced Plasmonic Architectures. Chemical Reviews.
- 634 **118** (6), 3032-3053 (2018).
- 635 26. Bathe, M., Rothemund, M. DNA Nanotechnology: A Foundation for Programmable
- 636 Nanoscale Materials. MRS Bulletin. 42 (12), 882-888 (2017)
- 637 27. Pilo-Pais, M., Acuna, G. P., Tinnefeld P., Liedl, T. Sculpting light by arranging optical
- components with DNA nanostructures. MRS Bulletin. 42 (12), 936-942 (2017).

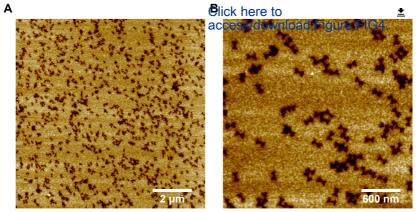
- 639 28. Graugnard, E., Hughes, W. L., Jungmann, R., Kostiainen, M. A., Linko, V. Nanometrology
- and super-resolution imaging with DNA. MRS Bulletin. 42 (12), 951-959 (2017).
- 29. Zhong, J. et al. Metallized DNA nanolithography for encoding and transferring spatial
- information for graphene patterning. *Nature Communications*. **4**, 1663 (2013).
- 30. Zhang, G., Surwade, S. P., Zhou, F., Liu, H. DNA nanostructure meets nanofabrication.
- 644 *Chemical Society Reviews.* **42** (7), 2488-2496 (2013).
- 645 31. Praetorius, F., Kick, B., Behler, K. L., Honemann, M. N., Weuster-Botz, D., Dietz, H.
- Biotechnological mass production of DNA origami. *Nature*. **552** (7683) 84-87 (2017).
- 647 32. Linko, V. et al. One-step large-scale deposition of salt-free DNA origami nanostructures.
- 648 Scientific Reports. **5**, 15634 (2015).
- 649 33. Arbabi A., Horie, Y., Bagheri, M., Faraon, A. Dielectric metasurfaces for complete control
- of phase and polarization with subwavelength spatial resolution and high transmission. *Nature*
- 651 *Nanotechnology.* **10** (11), 937-943 (2015).
- 652 34. Shen, B. et al. Plasmonic nanostructures through DNA-assisted lithography. *Science*
- 653 Advances. 4 (2), eaap8978 (2018).
- 654 35. Douglas, S. M. et al. Rapid prototyping of 3D DNA-origami shapes with caDNAno. *Nucleic*
- 655 Acids Research. **37** (26), 5001-5006 (2009).
- 656 36. Ke, Y. et al. Multilayer DNA Origami Packed on a Square Lattice. *Journal of the American*
- 657 *Chemical Society.* **131** (43), 15903-15908 (2009).
- 658 37. Dietz, H., Douglas, S. M., Shih, W. M. Folding DNA into twisted and curved nanoscale
- 659 shapes. Science. **325** (5941), 725-730 (2009).
- 660 38. Castro, C. E. et al. A primer to scaffolded DNA origami. *Nature Methods.* **8** (3), 221-229
- 661 (2011).
- 662 39. Kim, D.-N., Kilchherr, F. Dietz, H., Bathe, M. Quantitative prediction of 3D solution shape
- and flexibility of nucleic acid nanostructures. Nucleic Acids Research. 40 (7), 2862-2868 (2011).
- 664 40. Benson, E. et al. DNA rendering of polyhedral meshes at the nanoscale. *Nature*. **523**
- 665 (7561) 441-444 (2015).
- 666 41. Veneziano, R. et al. Designer nanoscale DNA assemblies programmed from the top
- 667 down. Science. **352** (6923), 1534 (2016).
- 668 42. Linko, V., Kostiainen, M. A. Automated design of DNA origami. Nature Biotechnology. 34
- 669 (8), 826-827 (2016).
- 670 43. Nummelin, S., Kommeri, J., Kostiainen, M. A., Linko, V. Evolution of Structural DNA
- 671 Nanotechnology. *Advanced Materials*. **30** (24), 1703721 (2018).
- 672 44. Maffeo, C., Yoo, J., Aksimentiev, A. De novo reconstruction of DNA origami structures
- 673 through atomistic molecular dynamics simulation. *Nucleic Acids Research.* **44** (7), 3013-3019
- 674 (2016).

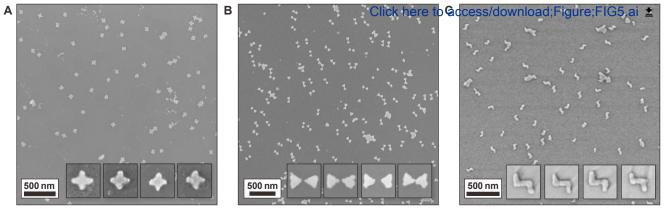
- 675 45. Stahl, E., Martin, T. G., Praetorius, F., Dietz, H. Facile and Scalable Preparation of Pure
- and Dense DNA Origami Solutions. Angewandte Chemie International Edition. 53 (47), 12735-
- 677 12740 (2014).
- 678 46. Shaw, A., Benson, E., Högberg, B. Purification of Functionalized DNA Origami
- 679 Nanostructures. ACS Nano. 9 (5), 4968-4975 (2015).
- 680 47. Kuzyk, A., Yurke, B., Toppari, J. J., Linko, V., Törmä, P. Dielectrophoretic Trapping of DNA
- 681 Origami. *Small.* **4** (4), 447-450 (2008).
- 48. Lin, C., Perrault, S. D., Kwak, M., Graf, F., Shih, W. M. Purification of DNA-origami
- 683 nanostructures by rate-zonal centrifugation. *Nucleic Acids Research*. 41 (2), e40 (2013).
- 684 49. Douglas, S. M. et al. Self-assembly of DNA into nanoscale three-dimensional shapes.
- 685 *Nature* **459** (7245), 414-418 (2009).
- 686 50. Ramakrishnan, S., Ijäs, H., Linko, V., Keller A. Structural stability of DNA origami
- 687 nanostructures under application-specific conditions. *Computational and Structural*
- 688 Biotechnology Journal. **16**, 342-349 (2018).
- 689 51. Kielar, C. et al. On the Stability of DNA Origami Nanostructures in Low-Magnesium
- 690 Buffers. Angewandte Chemie International Edition. 57 (30), 9470-9474 (2018).
- 691 52. Surwade, S. P. et al. Nanoscale growth and patterning of inorganic oxides using DNA
- 692 nanostructure templates. Journal of the American Chemical Society. 135 (18), 6778-6781
- 693 (2013).
- 694 53. Shen, B., Linko, V., Tapio, K., Kostiainen, M. A., Toppari, J. J. Custom-shaped metal
- 695 nanostructures based on DNA origami silhouettes. *Nanoscale*. **7** (26), 11267-11272 (2015).
- 696 54. Liu, W., Zhong, H., Wang, R., Seeman, N. C. Crystalline Two-Dimensional DNA-Origami
- 697 Arrays. Angewandte Chemie International Edition. **50** (1), 264-267 (2011).
- 698 55. Shen, B., Kostiainen, M. A., Linko, V. DNA Origami Nanophotonics and Plasmonics at
- 699 Interfaces. Langmuir. **34** (49), 14911-14920 (2018).

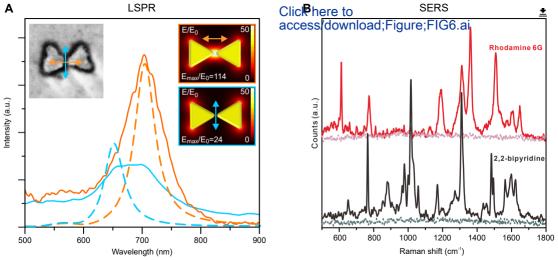












# Folding buffer (FOB) component concentrations [mM]

	Tris	Acetic acid	EDTA	Magnesium chloride	рΗ
2.5x FOB	100	47.5	2.5	31.25	~8,3
1x FOB	40	19	1	12.5	~8,3

Temperature range [ $^{\circ}$ C] Cooling rate 90-70 -0.2  $^{\circ}$ C / 8 s 70-60 -0.1  $^{\circ}$ C / 8 s 60-27 -0.1  $^{\circ}$ C / 2 s Hold until stopped

# PECVD and RIE parameters

	Gas	Gas flow [sccm]	Chamber pressure [mTorr]
PECVD of a-Si	5% SiH <sub>4</sub> in N <sub>2</sub>	500	1000
O <sub>2</sub> plasma treatment	$O_2$	50	40
RIE of SiO <sub>2</sub>	CHF <sub>3</sub>	25	
	Ar	25	30
RIE of a-Si	$O_2$	8	
	SF <sub>6</sub>	100	90
RIE of remaining a-Si	$O_2$	8	
	SF <sub>6</sub>	100	90

RF power [W] 15	Temperature [°C] 250	Duration [s] 90
200	30	1200
100	25	10-22
50	30	35
50	30	35-40

# Name of Material/ Equipment

Acetone

Agarose

Ammonium hydroxide

**BRANSON 5510** 

**Dimension Icon** 

Electron-beam evaporator IM-9912

Ethidium bromide

Eon Microplate spectrophotometer

Gel Doc XR+ Documentation System

Gel Loading Dye, Blue (6×)

G-storm GS1 Thermal cycler

HBR 4

Hydrofluoric acid

Isopropanol

Magnesium chloride

Mini-Sub Cell GT Horizontal Electrophoresis System

Plasmalab 80+ PECVD

Plasmalab 80+ RIE

Poly(ethylene glycol)

PowerPac HC High-Current Power Supply

Sapphire substrate (Al2O3)

Sigma VP

Silica gel

Single-stranded Scaffold DNA, type p7249

Sodium chloride

Staple strands (oligonucleotides)

TAE buffer (50×) pH 8.0

Take3 micro-volume plate

Tetraethyl orthosilicate

Company	Catalog Number

Honeywell 40289H Fisher Bioreagents 1036603 Fisher Chemical 10652251

Branson Bruker

Instrumentti Mattila

Sigma Aldrich E8751

BioTek BioRad

New England Biolabs B7021S

**Gene Technologies** 

IKA

Honeywell 40213H Honeywell 40301H Sigma Aldrich M8266

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Merck 1019691000

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**Integrated DNA Technologies** 

VWR Chemicals 444125D

BioTek

Sigma Aldrich 86578

# **Comments/Description**

Semiconductor grade ULSI, ≥ 99.5 %

Low-EEO, multi-purpose and molecular biology grade

25 % ammonia solution, Certified AR for Analysis, d = 0.91

Ultrasonic bath

Atomic force microscope

**Evaporator for PVD** 

Fluorescent dye for DNA staining

UV/Vis spectrophotometer used for DNA origami concentration measurements

Gel imaging system

Bromophenol blue-based loading dye for agarose gel electrophoresis

Heating bath

Semiconductor grade, 49.5-50.5 %

Semiconductor grade VLSI, ≥ 99.8 %

Anhydrous, ≥ 98 %

PECVD system

RIE system

BioUltra, 8,000

Thickness: 430 µm, Polish: DSP, Size: 50.8 mm

Scanning electron microscope

With indicator (orange gel), granulate ~1-3 mm

At 100 nM concentration

ACS reagent, ≥ 99.0 %

Sequences can be ordered e.g. at 100 micromolar in Rnase-free water

Electran Electrophoresis grade

Used for DNA origami concentration measurements

≥ 99.0 % (GC)



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#### POINT-TO-POINT RESPONSE TO EDITORS

26.06.2019

We wish to thank the editor for the additional comments on our manuscript "DNA Origami–Mediated Substrate Nanopatterning of Inorganic Structures for Sensing Applications" (Manuscript ID: JoVE60313R1).

Please find our detailed response to the comments below. We hope that the enclosed revised manuscript (with tracked changes) supported with the new files answers all the minor comments put forth by the editor, and that the manuscript could now be published in this form in *Journal of Visualized Experiments* – please do not hesitate to get in touch if you have any further queries.

Yours sincerely.

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#### **Editorial comments:**

1. The caDNAno supplemental file is missing. Please provide the sequences of the plasmid and the staple strands in a Table.

We apologize that overshooting. We have now uploaded the caDNAno file (.json), all staple sequences (.xlsx) and the sequence of M13mp18 (fasta file .txt).

2. Please revise the highlighting of the protocol to be under 2.75 pages. Additional, many highlighted steps lack the required detail to film. Please include the granular details in the highlighting for the video as these parameters are needed. For example, step 2.3 needs the temperature ramp included in the highlighting for filming.

We have revised the highlighting and paid extra attention to the details in each step. To avoid unnecessary listing of parameters for the highlighting, we have transfer long lists of parameters from the text to Tables (the termal ramp, composition of FOB and etching/deposition parameters as .xlsx). The length of the highlighted part (without line breaks) is now 2.5 pages.

# 3. Please ensure that this video highlighting tells a complete story for continuity.

We have carefully went through the highlighted parts and in our opinion the parts form a complete story. The sections that do not include any highlighting are optional or related to final sample characterization (not crucial for the protocol).

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Supplemental File 1

Click here to access/download **Supplemental Coding Files**bowtie\_poly-T.json

Supplemental File 2

Click here to access/download **Supplemental Coding Files**m13mp18\_sequence.txt

Supplemental File 3

Click here to access/download **Supplemental Coding Files**bowtie\_staple\_sequences.xlsx