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## Growth of Gold Dendritic Nanoforests on Titanium Nitride-Coated Silicon Substrates --Manuscript Draft--

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**TITLE:**

Growth of Gold Dendritic Nanoforests on Titanium Nitride-Coated Silicon Substrates

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**KEYWORDS:**

nano, gold, titanium nitride, silicon, galvanic replacement reaction, dendrite, synthesis

**SUMMARY:**

This study presents a feasible procedure for synthesizing gold dendritic nanoforests on titanium nitride/silicon substrates. The thickness of gold dendritic nanoforests increases linearly within 15 min of a synthesis reaction.

**ABSTRACT:**

In this study, a high-power impulse magnetron sputtering system is used to coat a flat and firm titanium nitride (TiN) film on silicon (Si) wafers, and a fluoride-assisted galvanic replacement reaction (FAGRR) is employed for the rapid and easy deposition of gold dendritic nanoforests (Au DNFs) on the TiN/Si substrates. Scanning electron microscopy (SEM) images and energy-dispersive X-ray spectroscopy patterns of TiN/Si and Au DNFs/TiN/Si samples validate that the synthesis process is accurately controlled. Under the reaction conditions in this study, the thickness of the Au DNFs increases linearly to  $5.10 \pm 0.20 \mu\text{m}$  within 15 min of the reaction. Therefore, the employed synthesis procedure is a simple and rapid approach for preparing Au DNFs/TiN/Si composites.

**INTRODUCTION:**

Gold nanoparticles have characteristic optical properties and localized surface plasmon resonances (LSPRs), depending on the size and shape of the nanoparticles<sup>1-4</sup>. Moreover, gold nanoparticles can significantly enhance plasmonic photocatalytic reactions<sup>5</sup>. Dendritic nanoforests stacked using gold nanoparticles have received considerable attention because of their noteworthy specific surface areas and robust LSPR enhancement<sup>6-13</sup>.

TiN is an extremely hard ceramic material and has remarkable thermal, chemical, and mechanical stability. TiN has distinctive optical properties and can be used for plasmonic applications with visible-to-near-infrared light<sup>14,15</sup>. Research has demonstrated that TiN can produce electromagnetic field enhancements, similar to Au nanostructures<sup>16</sup>. The deposition of copper<sup>17</sup> or silver<sup>18–20</sup> on TiN substrates for applications has been demonstrated. However, few studies have been performed on Au/TiN composite materials for applications. Shiao et al. have recently demonstrated potential applications of Au DNFs/TiN composites for photoelectrochemical cells<sup>21</sup> and chemical degradation<sup>22</sup>.

Au can be synthesized on a TiN substrate by using a FAGRR<sup>23</sup>. The deposition condition of Au DNFs on TiN is crucial in the performance of applications. This study examines the growth of Au DNFs on a TiN-coated Si substrate.

## **PROTOCOL:**

### **1) Sample preparation**

#### **1.1) TiN substrate preparation using a high-power impulse magnetron sputtering system**

1.1.1) Cut a 4 inch n-type silicon wafer into 2 cm x 2 cm samples.

1.1.2) Wash the samples using acetone, isopropanol, and deionized water.

1.1.3) Dry them using an N<sub>2</sub> spray for 5 min.

1.1.4) Place the washed Si samples in a sample holder and place the sample holder into a high-power impulse magnetron sputtering (HIPIMS) chamber.

1.1.5) Place a titanium target with a diameter of 4 inches on a sputtering cathode.

1.1.6) Reduce the chamber pressure to less than  $8 \times 10^{-6}$  Torr by using a mechanical pump and cryopump.

1.1.7) Use HIPIMS to deposit a Ti layer on a silicon wafer and deposit a TiN layer on the Ti layer. See **Table 1** for the deposition parameters of Ti and TiN layers in HIPIMS.

#### **1.2) Au DNF preparation on TiN/Si substrates**

1.2.1) Place 24 mL of a reactant solution comprising 10 mM chloroauric acid (HAuCl<sub>4</sub>) and buffered oxide etchant solution comprising 11.3% NH<sub>4</sub>F and 2.3% HF into a Teflon container measuring 5 cm x 5 cm x 5 cm.

1.2.2) Dip the substrates into the mixture solution for 3 min.

1.2.3) Remove the sample and wash it using deionized water.

1.2.4) Dry the sample using the N<sub>2</sub> spray and then incubate it at 120 °C for 5 min to obtain Au DNFs/TiN/Si samples.

1.2.5) Repeat the Au DNF preparation 10x.

## 2) Sample examination

### 2.1) Scanning electron microscopy analyses

2.1.1) Cut the sample into 0.4 cm x 0.8 cm with a tungsten pen.

2.1.2) Coat a thin Au film on the sample by an ion sputter coater for 50 s and clean it using the N<sub>2</sub> spray.

2.1.3) Place the prepared sample into a scanning electron microscopy (SEM) instrument.

2.1.4) Obtain SEM images by the scanning electron microscope and conduct element analysis<sup>21,22</sup>.

### 2.2) X-ray diffraction analyses

2.2.1) Place the sample into an X-ray diffraction (XRD) instrument.

2.2.2) Obtain XRD patterns<sup>21,22</sup>.

## REPRESENTATIVE RESULTS:

**Figure 1** depicts images of the Au DNFs/TiN/Si sample preparations. The silicon wafer was silvery white (**Figure 1a**). TiN/Si was golden yellow and had a homogeneous surface (**Figure 1b**), which indicated the uniform TiN coating on the silicon wafer. Au DNFs/TiN/Si was yellowish brown and less homogeneous on the surface (**Figure 1c**) because of the random distribution of Au DNFs.

**Figure 2** presents the plan and cross-sectional SEM images of Au DNFs deposited on the TiN/Si substrates. The TiN layer had a uniform surface (**Figure 2a**), and the thickness of the TiN layer was approximately 300 nm (**Figure 2b**). At 1 min, small Au nuclei were observed everywhere (**Figure 2c**), some of which developed into a large nucleus similar to a sea urchin (**Figure 2d**). A single tree-like structure was formed at 3 min (**Figure 2e,f**), and these branching were observed to overlap at 5 min (**Figure 2g,h**). At 10 min, Au DNFs formed and covered the entire TiN layer (**Figure 2i,j**). At 15 min, dense Au DNFs were formed (**Figure 2k**), and the thickness of the DNFs reached 5 μm (**Figure 2l**).

**Figure 3** shows the energy-dispersive X-ray spectroscopy (EDS) analysis results of TiN/Si and Au DNFs/TiN/Si. The indicated elements agreed with the synthesis procedure. Furthermore, the evident peaks could validate that the coating of TiN and the synthesis of Au DNFs were not polluted.

**Figure 4** illustrates the variation of the thickness of the Au DNFs on the TiN/Si substrate with FAGRR time. The thickness of the Au DNFs increased linearly with synthesis time. The linear

equation on the thickness and synthesis time, which ranged from 1 to 15 min, was expressed as follows:  $y = 0.296t + 0.649$ .

**Figure 5** shows the XRD patterns of samples obtained by different deposition times. A strong (111) orientation of Au peaks was identified. The sharp cubic Au patterns, Au(111), Au(200), Au(220), and Au(311), agreed with JCPDS 04-0784. The increase in Au peaks with the deposition time corresponded to the growth of Au DNFs on the TiN/Si substrate. On the other hand, TiN peaks, namely TiN(111), TiN(200), TiN(220), and TiN(311), were evident at 1 min deposition, agreeing with JCPDS 38-1420. After 1 min, the signals of TiN gradually disappeared because the TiN/Si substrate was gradually covered by Au DNFs. These XRD results corresponded to previous reports<sup>21,22</sup>.

#### FIGURE AND TABLE LEGENDS:

**Table 1: Conditions for preparing Ti and TiN.** HIPIMS parameters for the deposition of Ti and TiN layers on a silicon wafer.

**Figure 1: Sample appearance.** Preparation of a 2 cm x 2 cm sample of (a) a silicon wafer, (b) TiN/Si, and (c) Au DNFs/TiN/Si.

**Figure 2: SEM pictures of samples.** The SEM overhead and cross-section views of Au DNFs deposited on the TiN/Si substrates at (a and b) 0 min; (c and d) 1 min; (e and f) 3 min; (g and h) 5 min; (i and j) 10 min; (k and l) 15 min.

**Figure 3: Elemental analysis.** EDS spectrum of (a) TiN/Si and (b) Au DNFs/TiN/Si.

**Figure 4: Au DNF thickness.** The thickness of Au DNFs on TiN/Si substrate at various synthesis times ( $n = 10$ )

**Figure 5: XRD patterns of samples.** XRD patterns of Au DNFs on TiN/Si substrate at different synthesis times.

#### DISCUSSION:

In this study, Au DNFs with multiple branch sizes were decorated on the surface of TiN/Si by using FAGRR. The deposition of the Au DNFs could be directly identified by a significant change in color. The thickness of the Au DNFs on TiN/Si increased to  $5.10 \pm 0.20 \mu\text{m}$  within 15 min, and this increase in thickness can be expressed using the following linear equation:  $y = 0.296t + 0.649$ , where the time varied from 1 to 15 min.

In FAGRR, the metal deposition is affected by the composition and the pH of the solution<sup>23</sup>. The deposition rate increases with the density of substrate surface defects. The thickness of the TiN layer decreases as the replacement reaction time increases. It is easy to remove Au DNFs from the TiN/Si substrates if the thickness of Au DNFs is thick enough.

Galvanic displacement reaction is more favorable for a metal with a higher redox potential<sup>23</sup>. In this study, the proposed facile and rapid electroless deposition process provides a

feasible approach for preparing Au/TiN/Si composites which can be used as visible-light photocatalysts<sup>21,22</sup>. Using the same protocol, it is also possible to fabricate Au DNFs on other substrates, such as TiN/SiO<sub>2</sub>/Si, TiN/glass, TiN/ITO, and TiN/FTO, for applications in the future.

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#### DISCLOSURES

The authors have nothing to disclose.

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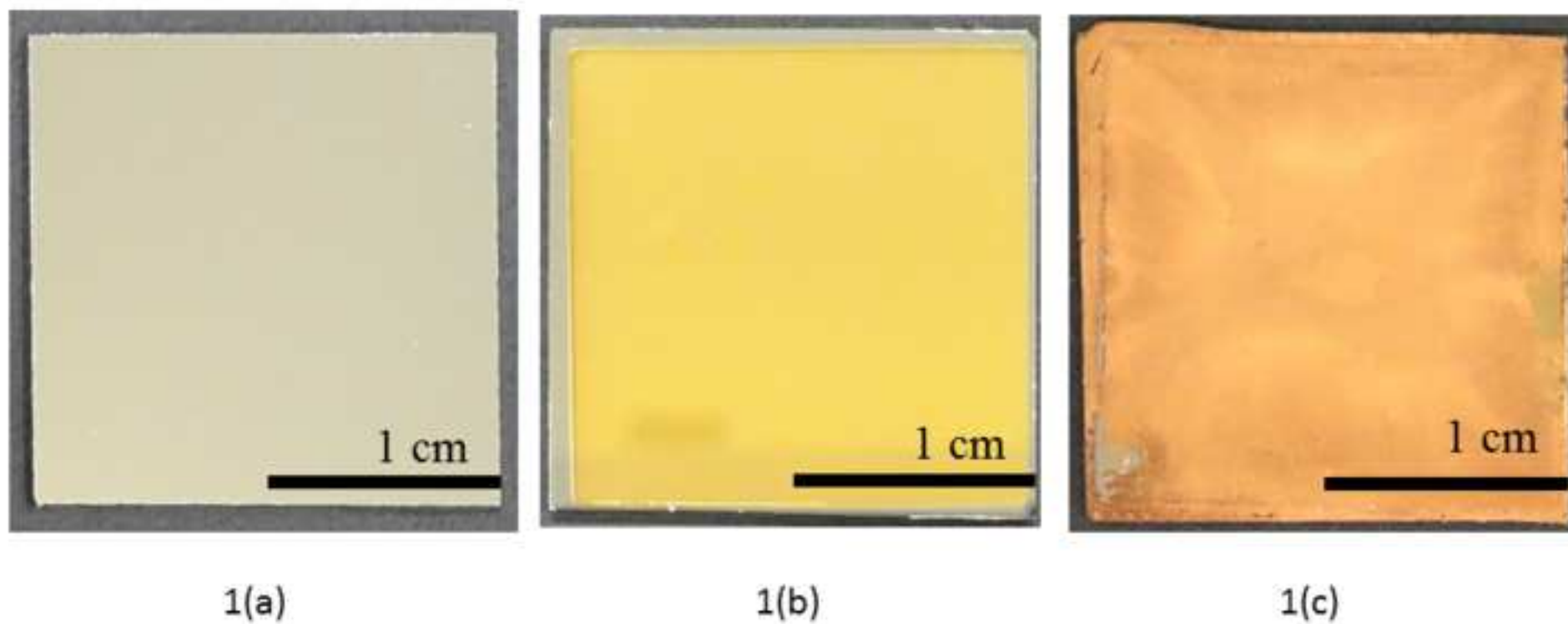
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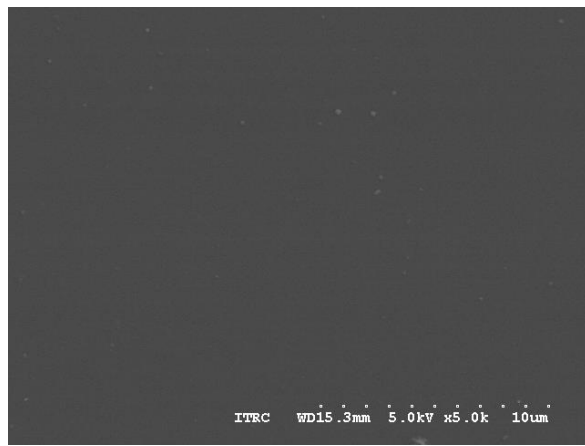
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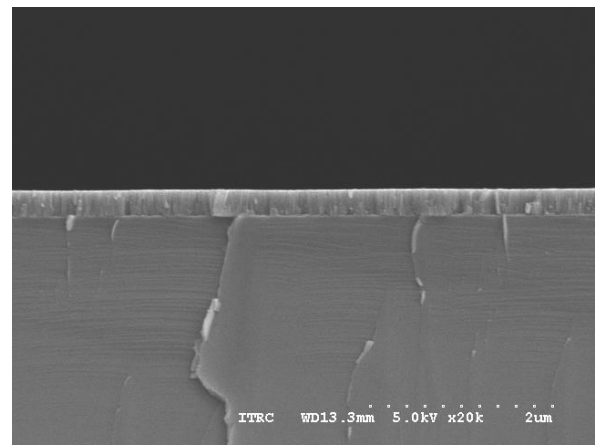
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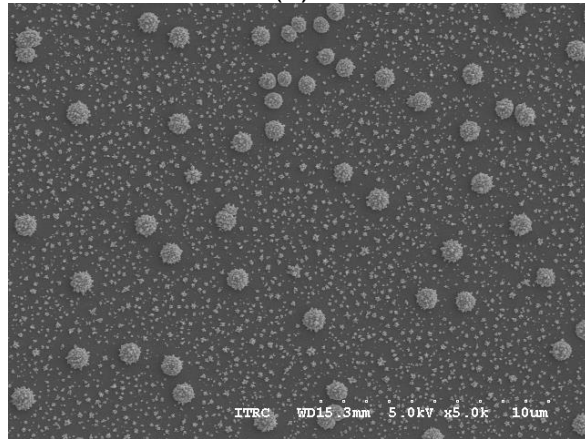




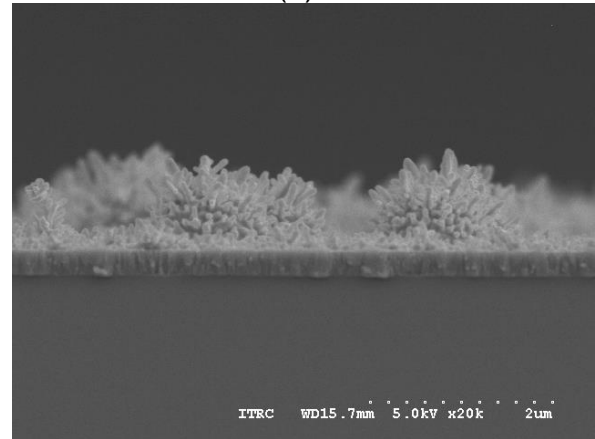
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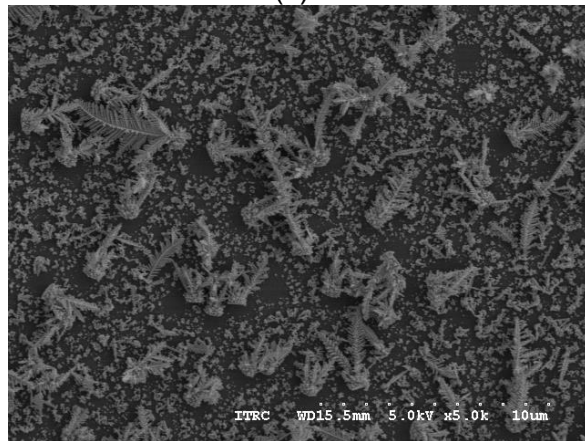
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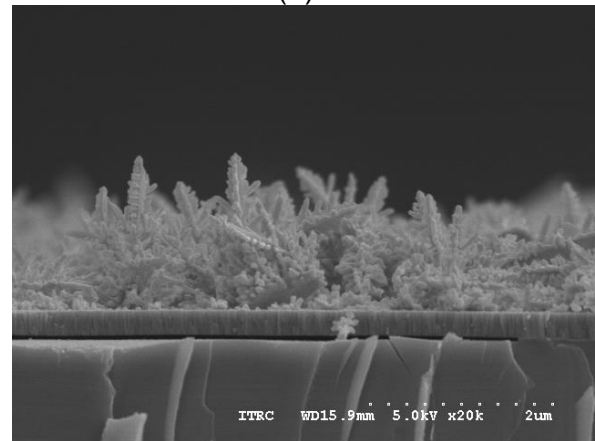
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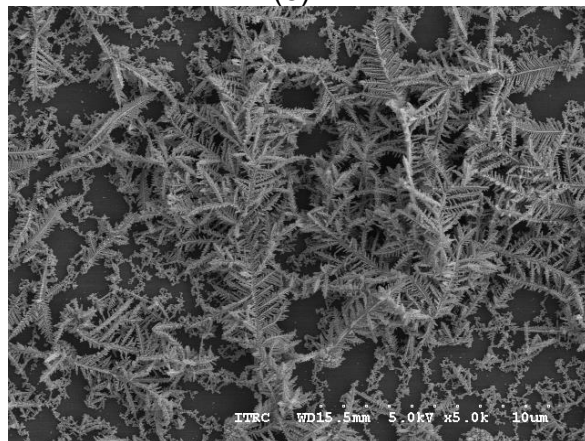
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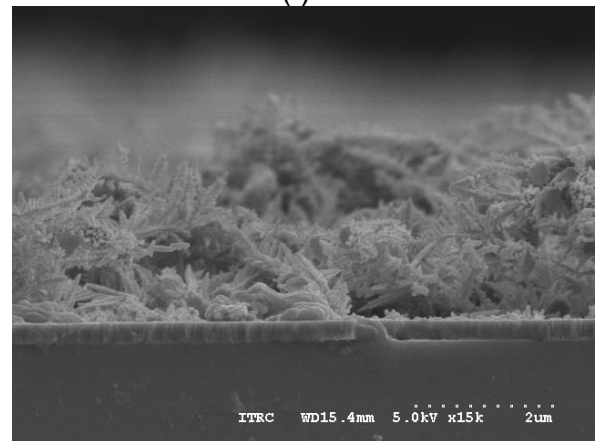
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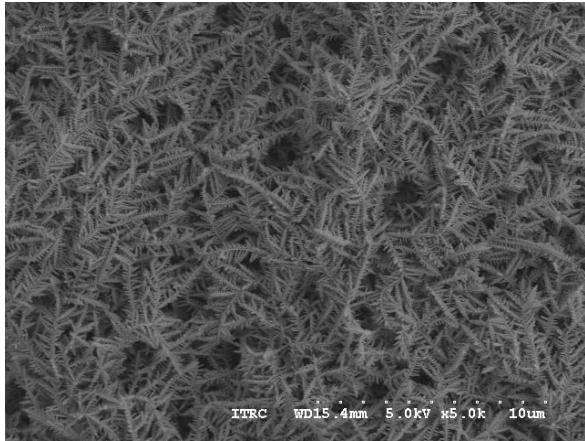
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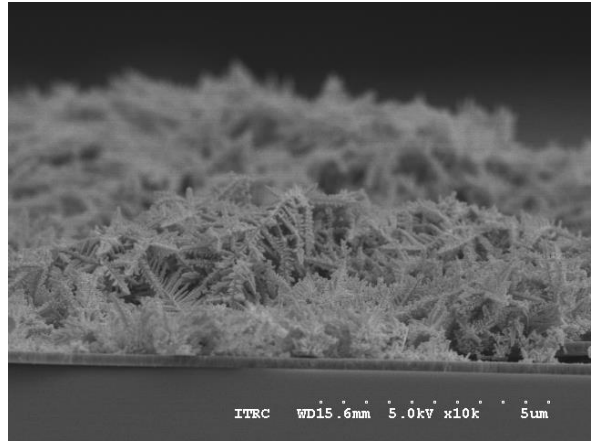
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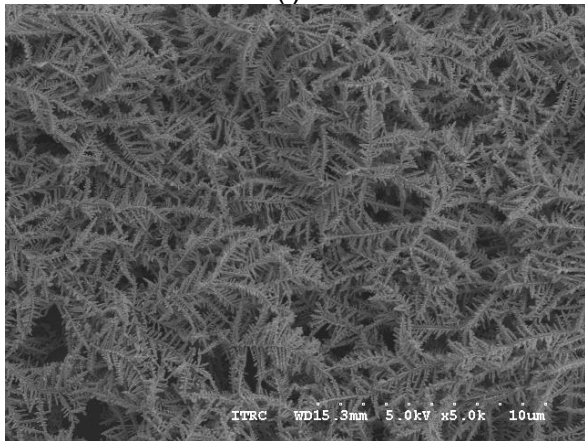
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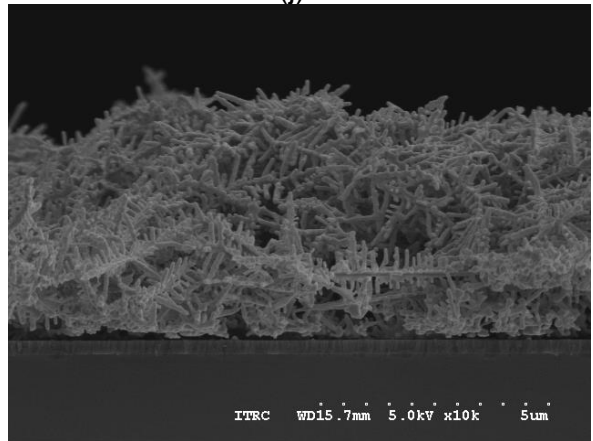
(i)



(j)



(k)



(l)

Figure 3

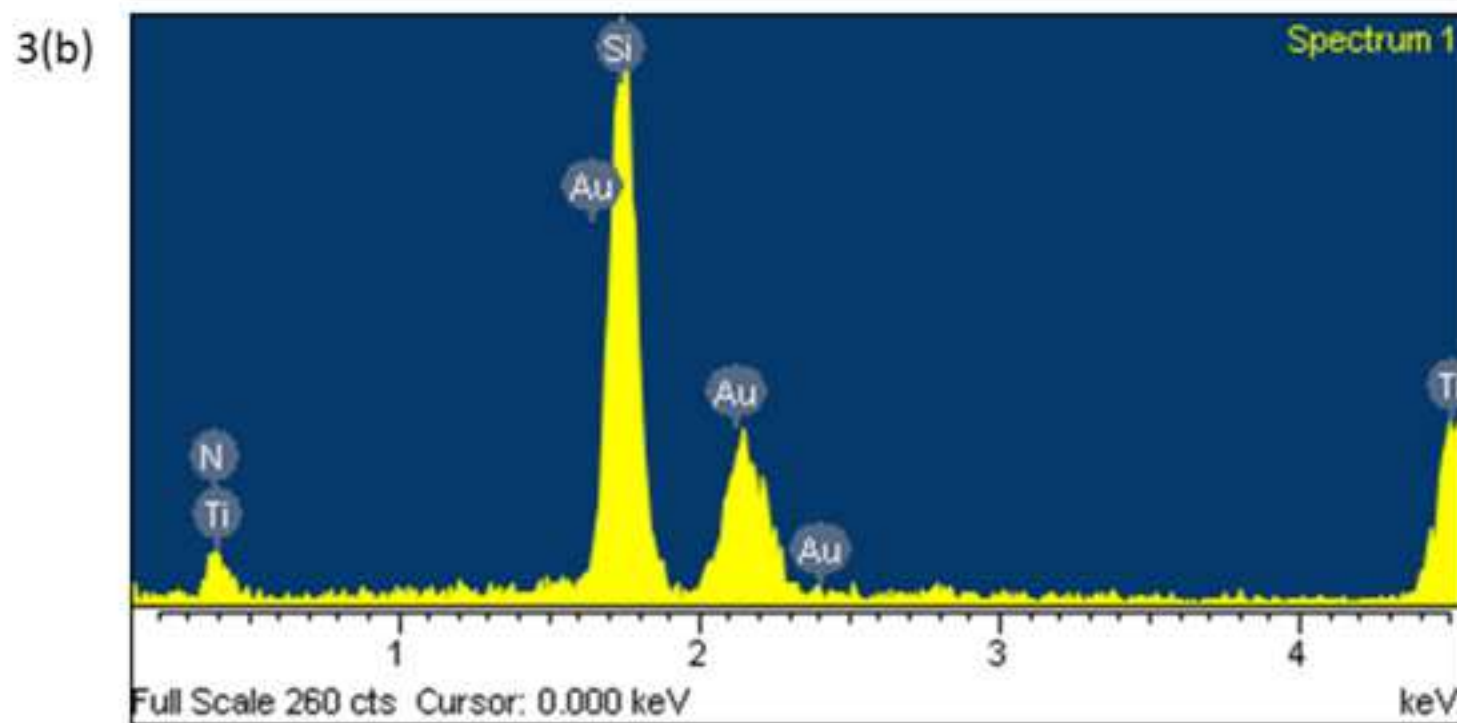
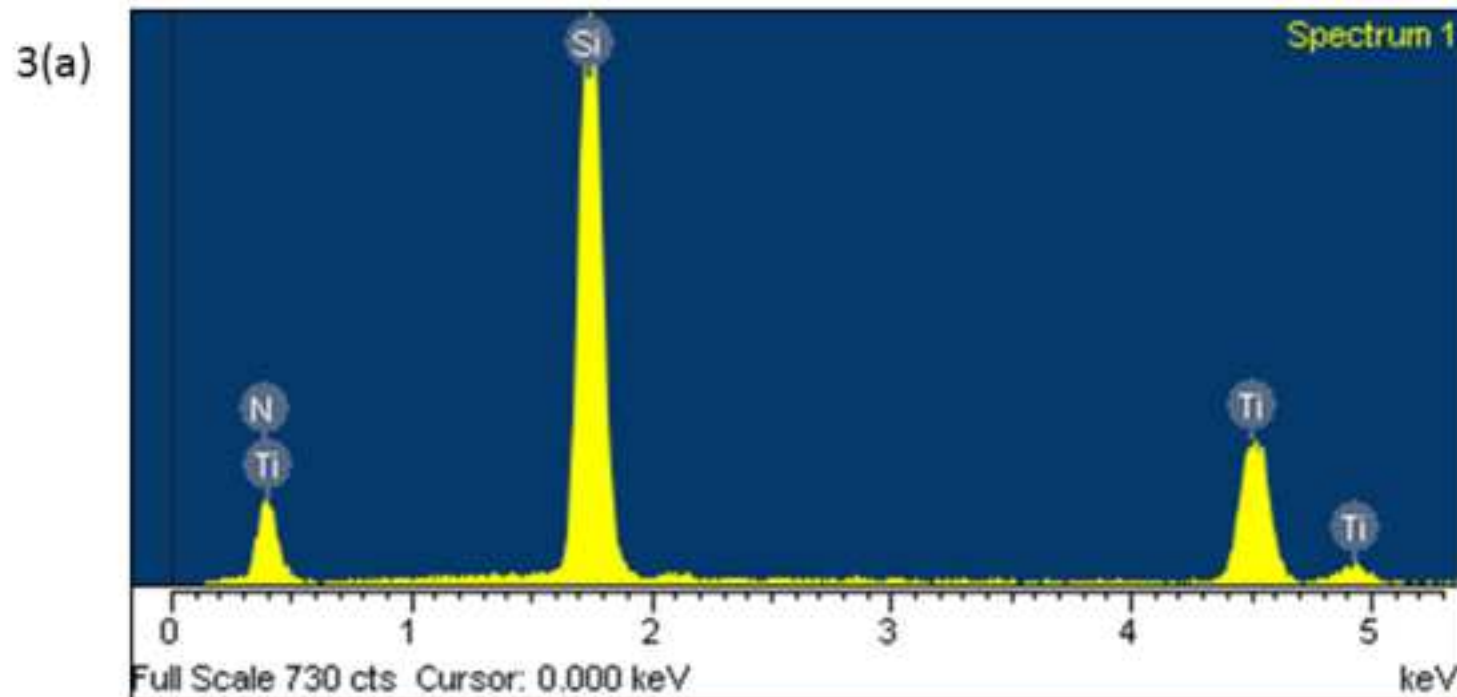


Figure 4

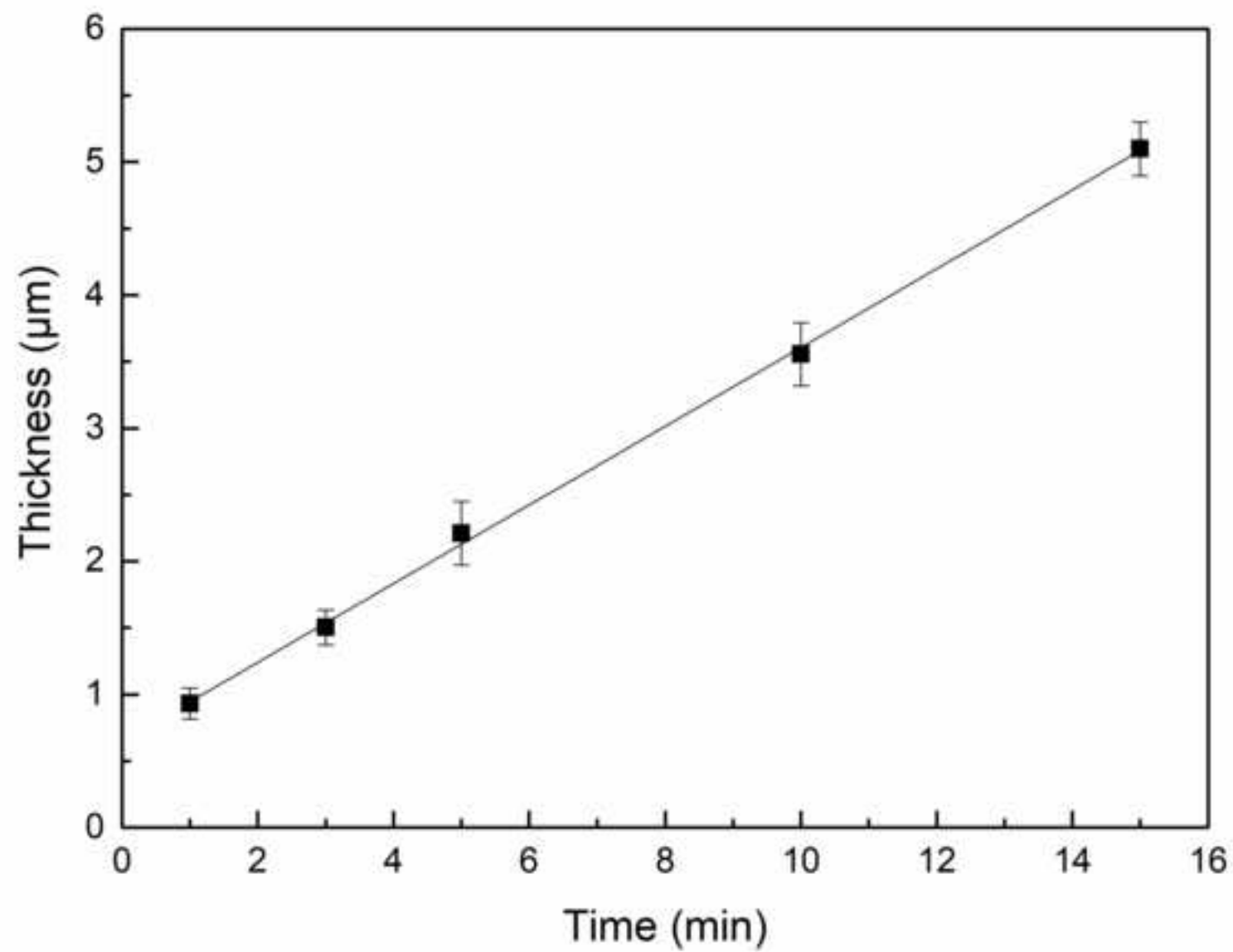
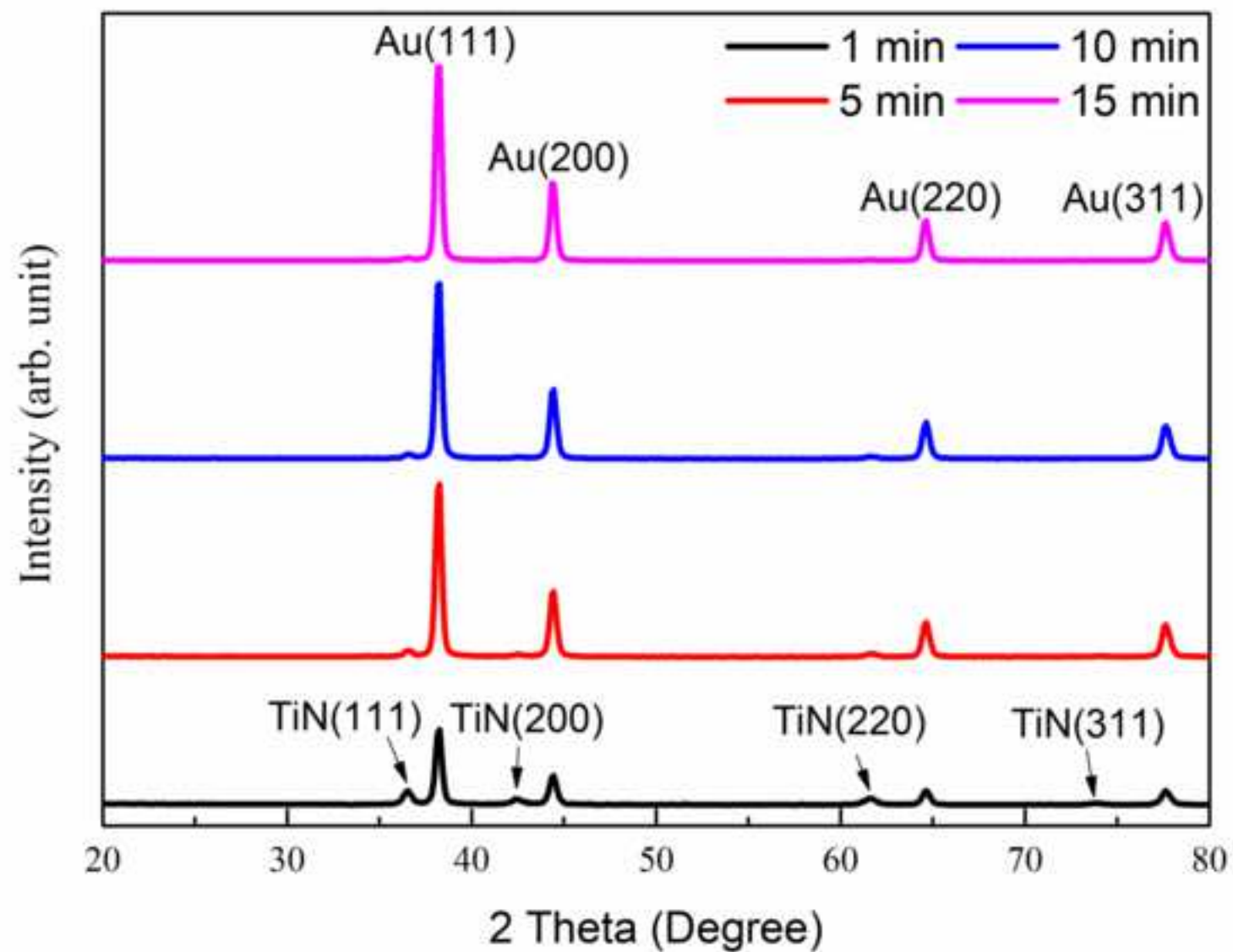


Figure 5



Substrate	DC power (W)	Impulse duration (μs)	Flow rate of Ar (sccm)	Flow rate of N <sub>2</sub> (sccm)
Ti layer	250	90	20	-
TiN layer	300	1000	30	1.5

Name of Material/Equipment	Company	Catalog Number
Acetone	Dinhaw Enterprise Co. Ltd.,Taipei, Taiwan	
Isopropanol	Echo Chemical Co. Ltd., Miaoli, Taiwan	TG-078-000000-75NL
Buffered Oxide Etch	Uni-onward Corp., Hsinchu, Taiwan	UR-BOE-1EA
Chloroauric Acid	Alfa Aesar., Heysham, United Kingdom	36400.03
N-Type Silicon Wafer	Summit-Tech Company, Hsinchu, Taiwan	
High-Power Impulse Magnetron Sputtering System (HiPIMS)	Melec GmbH, Germany	
Scanning Electron Microscope (SEM)	JEOL, Japan	
Ion Sputter Coater	Hitachi, Japan	
X-Ray Diffractometer (XRD)	PANalytical, The Netherlands	

Comments/Description
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Date:

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