

# Journal of Visualized Experiments

## Rapid collection of floral fragrance volatiles using a headspace volatile collection technique for GC-MS thermal desorption sample analysis --Manuscript Draft--

Article Type:	Invited Methods Article - JoVE Produced Video
Manuscript Number:	JoVE58928R3
Full Title:	Rapid collection of floral fragrance volatiles using a headspace volatile collection technique for GC-MS thermal desorption sample analysis
Section/Category:	JoVE Biology
Keywords:	charcoal filter; semiochemicals; portable; Vacuum; scent; flower
Corresponding Author:	Haleigh Ray University of Florida Gainesville, Florida UNITED STATES
Corresponding Author's Institution:	University of Florida
Corresponding Author E-Mail:	hray12@ufl.edu
Order of Authors:	Haleigh Ray Charles Stuhl Jennifer Gillett-Kaufman
Additional Information:	
Question	Response
Please indicate whether this article will be Standard Access or Open Access.	Standard Access (US\$2,400)
Please indicate the <b>city, state/province, and country</b> where this article will be <b>filmed</b> . Please do not use abbreviations.	Gainesville, Florida, USA

**TITLE:**

Rapid Collection of Floral Fragrance Volatiles using a Headspace Volatile Collection Technique for GC-MS Thermal Desorption Sampling

**AUTHORS & AFFILIATIONS:**

Haleigh A. Ray<sup>1</sup>, Charles J. Stuhl<sup>2</sup>, Jennifer L. Gillett-Kaufman<sup>1</sup>

<sup>1</sup>Entomology and Nematology Department, University of Florida, 1881 Natural Area Drive, Gainesville, Florida

<sup>2</sup>Center for Medical, Agricultural and Veterinary Entomology, Agricultural Research Service U.S. Department of Agriculture, Gainesville, FL

Haleigh A. Ray (haleighray12@gmail.com)

Charles J. Stuhl (charles.stuhl@ars.usda.gov)

Jennifer L. Gillett-Kaufman (gillett@ufl.edu)

**CORRESPONDING AUTHOR:**

Haleigh A. Ray (haleighray12@gmail.com)

**KEYWORDS:**

charcoal filter, semiochemicals, portable, vacuum, scent, flower

**SUMMARY:**

Here, we present a protocol for collecting the floral fragrance volatiles from blooming flowers, using a non-destructive sampling procedure.

**ABSTRACT:**

Fragrances of many flower families have been sampled and the volatiles analyzed. Knowing the compounds that make up the fragrances can be an important step to conservation of flowers that are threatened or endangered. Because floral fragrance is critical for attracting pollinators, this method could be used to better understand or even enhance pollination. We present a protocol using a portable charcoal air filter and vacuum to collect floral fragrance volatiles, which are then analyzed by a GC-MS. By using this method, fragrance volatiles can be sampled using a non-destructive method with a machine that is easily transported. This methodology uses a rapid sampling procedure, cutting sampling time down from 2-3 hours to approximately 10 minutes. Using GC-MS, the fragrance compounds can be identified individually, based on authentic standards. The steps used for collecting fragrance and control data are presented, from material setup to collecting the data output.

**INTRODUCTION:**

Flowers typically produce a fragrance used to attract pollinators. These fragrances are made up of many chemical compounds all acting together as a floral blend<sup>1-3</sup>. Without these fragrances, flowers would be less likely to pass on their genetic information using pollinators. Floral fragrance has been documented in many flowering plant families, with *Orchidaceae* being one of the more

common families studied<sup>4</sup>. To understand the role of floral fragrance in pollination, it is important to nondestructively collect and analyze the chemical compounds being emitted from the flowers at different times of the day and during the several days to weeks the flower blooms are open, as fragrance can vary over time<sup>5</sup>.

An early protocol for this kind of sampling was developed by Heath and Manukian<sup>6</sup>. The goal of their sampling methods was to reduce stress on the specimen (e.g., plants, insects) being studied. Earlier papers documented that destructive procedures to the plant were required, such as removing blooming flowers in order to collect the fragrance. More recent floral fragrance publications by Cancino and Damon<sup>7,8</sup> used similar methods. This study put the flowers in glass chambers and passed purified air over them; then fragrance compounds from the chamber were absorbed onto porous polymer adsorbents in clear Pasteur pipettes. The fragrances were collected for at least two hours during this study. Sadler et al.<sup>9</sup> carried out floral fragrance studies on an epiphytic orchid in south Florida, much like the original study<sup>10</sup>. Again, this study required the flowers to be sampled for over two hours to collect the fragrance volatiles, with fragrance collected onto the porous polymer adsorbent. The paper here presents a non-destructive method that allows for much quicker sampling, lasting only 10 minutes. Also, instead of using a glass chamber oven baking bags are used, which allow for more flexible movement of the chamber and reduce the chances of damage to the flowers. These bags come in several sizes allowing the option to select the size of bag that will easily fit individual samples without damaging the sample or the surrounding material. The adsorbent used in this study was Tenax Porous Polymer Adsorbent. This differs from Porapak, because the sample can be thermally desorbed onto the GC-MS column for analysis, eliminating the use of a chemical solvent.

The methods in this study provide a way to quickly sample fragrance volatiles produced by flowers and could be used to sample volatiles from other specimens as well, such as insect pheromones, or mushroom volatiles. The reduced time for sampling means there is less stress on the sample and the ability to collect many samples in a short period of time. For example, in Sadler et al.<sup>9</sup>, the flower was only fragrant at night, so only two or three samples could be collected each night. With the method here, samples could be taken all night at 15-20-minute intervals from the same flower. Additionally, by using bags instead of glass chambers, the headspace can be suspended easier for sampling in the field for in situ collection on endangered or threatened plant species. Using the method presented here, we were able to sample flowers 1.5 to 2 meters above ground. These methods are incredibly useful for fragrance collection in the laboratory and the field, and provides researchers with a sampling technique that is fast and non-destructive to the sample.

## **PROTOCOL:**

NOTE: Perfumes or scented lotions and products must not be worn during any of these procedures.

### **1. Flower selection**

NOTE: Flowers used can be either naturally growing in the environment or kept under artificial environmental conditions. Temperature, humidity, and light level during collection can vary based on the specific flower species used, and what type of data is being collected. For example, data has been collected during the day and at night for the same flower to determine if fragrance varies over time of day, and collected from both in situ and greenhouse flowers.

1.1. Select a flower that is initially unopened, to standardize sample collection time. This controls for a flower changing fragrance over time.

1.2. Depending on the duration of the blooming time, if possible, wait at least 24 hours after blooming to collect the sample, setting a standard time for all samples.

1.3. If there are multiple blooming flowers on a plant, mark the one that will be used with flagging tape or something similar to ensure repeated sampling of the same flower.

## 2. Material preparation

2.1. Use oven bags (approximately 40.5 cm × 44.5 cm) and corrugated PTFE tubing.

2.2. Initially, boil oven bags in water for ~30 min to remove residual plastic compounds. To dry, bake in an oven at 175 °C.

2.3. Once the bags have dried, add a polypropylene bulkhead union to each corner of the closed end of the oven bags. These attachments allow connection of tubes to push charcoal filtered air in and pull fragrance out of the headspace.

2.4. Rinse all bags and tubes with 75% ethanol. Let both air dry after rinsing.

2.5. After the oven bags have dried, bake bags and tubes in an oven at low heat, approximately 74-85 °C for 30 min.

## 3. Volatile collection

NOTE: Sterile neoprene gloves need to be worn throughout this process, as contacting the bag or filter cartridges can contaminate the samples.

3.1. Cover the selected flower with a baked oven bag. Cinch the bag together tightly with a plastic zip tie below the flower to prevent unwanted airflow into the bag.

3.2. Attach a tube from the air outlet of the collection equipment and connect it to one of the bulkhead unions on the oven bag.

3.3. On the other bulkhead union, attach a glass filter cartridge containing porous polymer adsorbent.

133  
134 3.4. Attach a second tube to the collection equipment on the vacuum input. Connect end of  
135 the second tube onto the glass volatile collection filter cartridge.

136  
137 3.5. Turn on both the air pump and vacuum at the same time set at ~0.05 L/min. The  
138 headspace around the flower will fill with air, but not become overinflated. The system will pull  
139 air from the bag through the filter, trapping the floral volatiles.

140  
141 3.6. Allow the machine to run for 10 min and then turn off both the air pump and vacuum.

142  
143 NOTE: Flower species that produce/emit a smaller amount of fragrance may need to be sampled  
144 for a longer period of time.

145  
146 3.7. Disassemble the tubes and the glass filter cartridge. Place the filter into a glass vial with a  
147 screw-on cap. Once the cap is on, seal the vial with PTFE pipe thread tape.

148  
149 3.8. Store samples in a freezer (-80 °C) until analyzed using GC-MS.

150  
151 3.9. Repeat this process with a clean oven bag and glass filter, this time with an empty oven  
152 bag, to collect a blank air sample as a control. This allows any background volatiles collected to  
153 be identified.

154  
155 NOTE: Repeating sample collection needs to be done at approximately the same time each day,  
156 as some flowers produce varying fragrance levels over the course of a day.

#### 157 158 4. **GC-MS**

159  
160 4.1. Remove the glass filter cartridge from the freezer and place into a GC-MS in the injector  
161 port.

162  
163 4.2. Release headspace volatiles collected on porous polymer adsorbent from the adsorbent  
164 by heating in the thermal collection trap (TCT) to 220 °C for 8 min within a flow of helium gas  
165 (rate: 1.2 mL/min).

166  
167 4.3. Collect desorbed compounds in the TCT cold trap unit at -130 °C. The cold trap  
168 temperature is regulated by the GC-MS program.

169  
170 4.4. Flash heat the TCT cold trap unit to inject the compounds into the capillary column of the  
171 gas chromatograph to which the TCT cold trap unit was connected. The method for the TCT  
172 begins at -20 °C and ends at 150 °C.

173  
174 4.5. Program the GC-MS to rise from 40 °C to 280 °C at 15 °C/min, with a 5 min hold at 40 °C.

#### 175 176 5. **Data analysis**

177  
178 5.1. For identification, compare the mass spectra of the sample to those from mass spectra  
179 libraries (NIST and Department of Chemical Ecology, Goteborg University, Sweden<sup>11</sup>), as well as  
180 retention times of the volatiles to times of authentic compound standards<sup>12</sup>.

181  
182 5.2. Compare chromatograms of collected volatiles to identify common reoccurring peaks.

183  
184 5.3. After identifying the peak volatiles, use Pherobase (online database of semiochemicals  
185 and pheromones) to determine if they have been previously described in floral fragrances<sup>10</sup>.

## 186 187 **REPRESENTATIVE RESULTS:**

188 Representative data from the GC-MS are shown as a chromatogram in **Figure 1**. In addition to  
189 the chromatogram, a data file of results is also provided (**Supplementary File 1**). This data file  
190 provides the retention time for each peak (RT), and an identification of what compound that peak  
191 is (Library/ID). Peaks between 10:00 and 15:00 minutes are floral volatiles, due to the molecular  
192 weight of the compounds<sup>10</sup>. The numbers above the peaks signify the retention times of the  
193 identified compounds which are referenced the data file of the results (**Supplementary File 1**).  
194 By obtaining the chromatogram and data file for each fragrance sample, the compounds can be  
195 compared and those that are reoccurring for each flower sample can be identified. Collections  
196 can be identified from this document under the category "Sample", named to represent the  
197 flower sampled, and the time and date of the collection (example: UF1 8AM 03/16/15). Page 1  
198 of this document also shows the identification of specific compounds identified from the sample  
199 (LibraryID), which peak retention time from **Figure 1** the compound corresponds to (Pk#), and  
200 the percentage of the total fragrance that each volatile comprises (Area %). All the collected  
201 volatiles listed under "Library/ID" can be referenced in Pherobase to determine if they have  
202 previously described in a floral fragrance. For example, in the **Supplementary File 1**, compound  
203 #21, with a retention time (RT) of 10.311 has been identified as benzaldehyde. In future samples,  
204 if benzaldehyde is present, it can be referenced on Pherobase to determine if it is a likely floral  
205 compound for the flower. In **Figure 2**, benzaldehyde was searched on Pherobase. Once a  
206 compound has been selected, the page shows a list of all flower species, organized by plant  
207 family, from which that fragrance compound has been identified. Highlighted in the bottom right  
208 corner of **Figure 2** is a small subset of the orchid species (Orchidaceae) from which benzaldehyde  
209 has been determined to be present in the floral fragrance.

## 210 211 **Figure Legends:**

212  
213 **Figure 1. GC-MS volatile peak results.** Graphical results showing the peak volatiles of the floral  
214 fragrance sample. Numbers above the peaks correspond to a list of all collected volatile  
215 compounds, identifying the peak to the specific volatile. Peaks between 10:00 and 15:00 minutes  
216 are most likely to be volatiles from a floral fragrance.

217  
218 **Figure 2. Pherobase example results.** An example of results from a Pherobase search for a  
219 fragrance compound. In this figure Benzaldehyde was searched, and the results show a list of all  
220 flower species from which this fragrance has been identified.

**DISCUSSION:**

Though this technique is incredibly valuable for its sampling speed and portability, one limitation is using it for epiphytic species, or those growing on trees and not from the ground. In the original study<sup>10</sup>, one of the flowers sampled was epiphytic. Because the machine is too heavy to hang freely, a stable, elevated base must be made for sampling. Additionally, the machine can either be plugged in to an electrical outlet or battery powered, so if there is prolonged field sampling, there must be a power source to charge the batteries when the machine is not in use.

The methods here allow for an in situ non-destructive sampling, with rapid repeated sampling and a much quicker sampling time. While some floral fragrance studies require the fragrance to be collected for 2-3 hours for one sample, the presented method can accurately collect the volatiles in approximately 10 min due to the collection material (porous polymer adsorbent) used in the glass filter.

These collection methods provide a way to quickly and safely sample the fragrance produced by flowers, without destroying or harming the flower. With so many flowers, especially those in the family *Orchidaceae*, being categorized as threatened or endangered, analyzing the fragrances they are producing in a non-destructive manner is critical as work is conducted to understanding their pollination biology. The information gained from these studies could potentially be used to boost pollination using synthetic blends based on peak chemicals found to attract more pollinators to areas with blooming orchids.

**DISCLOSURES:**

The authors declare no conflicts of interest.

**ACKNOWLEDGEMENTS:**

USDA-ARS Research Project number 6036-22000-028-00D. The use of trade, firm, or corporation names in this publication is for the information and convenience of the reader. Such use does not constitute an official endorsement or approval by the United States Department of Agriculture or the Agriculture Research Service of any product or service to the exclusion of others that may be suitable. Additionally, the University of Florida Biology Department- Lewis and Varina Vaughn Fellowship in Orchid Biology (2017), and a University of Florida Graduate Research Fellowship (2014-2018) provided funding as well.

**REFERENCES:**

- 1 Knudsen, J.T., Tollsten, L., Bergstrom, L.G. Floral scents- A checklist of volatile compounds isolated by head-space techniques. *Phytochemistry*. **33**, 253-280 (1993).
- 2 Dudareva, N.A., Pichersky, E. Biology of floral scent. Boca Raton, FL: CRC Taylor and Francis. (2006).
- 3 Altenburger, R., Matile, P. Rhythms of fragrance emission in flowers. *Planta*. **174**, 242-247 (1988).

- 4 Dodson, C.H., Dressler, R.L., Hills, H.G., Adams, R.M., Williams, N.H. Biologically active compounds in orchid fragrances. *Science*. **164**, 1243-1249 (1969).
- 5 Theis, N., Raguso, R.A. The effect of pollination on floral fragrance in thistles. *Journal of Chemical Ecology*. **31** (11), 2581-2600 (2005).
- 6 Heath, R.R., Manukian, A. Development and evaluation of systems to collect volatile semiochemicals from insects and plants using a charcoal-infused medium for air purification. *Journal of Chemical Ecology*. **18**, 1209-1226 (1992).
- 7 Cancino, A., Damon, A. Comparison of floral fragrance components of species of *Encyclia* and *Prosthechea* (Orchidaceae) from Soconusco, southeast Mexico. *Lankesteriana*. **6**, 83-139 (2006).
- 8 Cancino, A., Damon, A. Fragrance analysis of euglossine bee pollinated orchids from Soconusco, south-east Mexico. *Plant Species Biology*. **22**, 129-134 (2007).
- 9 Sadler, J.J., Smith, J.M., Zettler, L.W., Alborn, H.T., Richardson, L.W. Fragrance composition of *Dendrophylax lindenii* (Orchidaceae) using a novel technique applied in situ. *European Journal of Environmental Science*. **1**, 137-141 (2011).
- 10 Ray, H.A., Stuhl, C.J., Gillett-Kaufman, J.L. Floral fragrance analysis of *Prosthechea cochleata* (Orchidaceae), an endangered native, epiphytic orchid, in Florida. *Plant Signaling and Behavior*. (2018).
- 11 National Institute of Standards and Technology. U.S. Department of Commerce. <https://www.nist.gov/>. Date Accessed: 16 July 2019.
- 12 The Pherobase: Database of Pheromones and Semiochemicals. <http://www.pherobase.com/> Date Accessed: 16 July 2019.



Figure 1

[Click here to access/download;Figure;Figure 1.png](#)

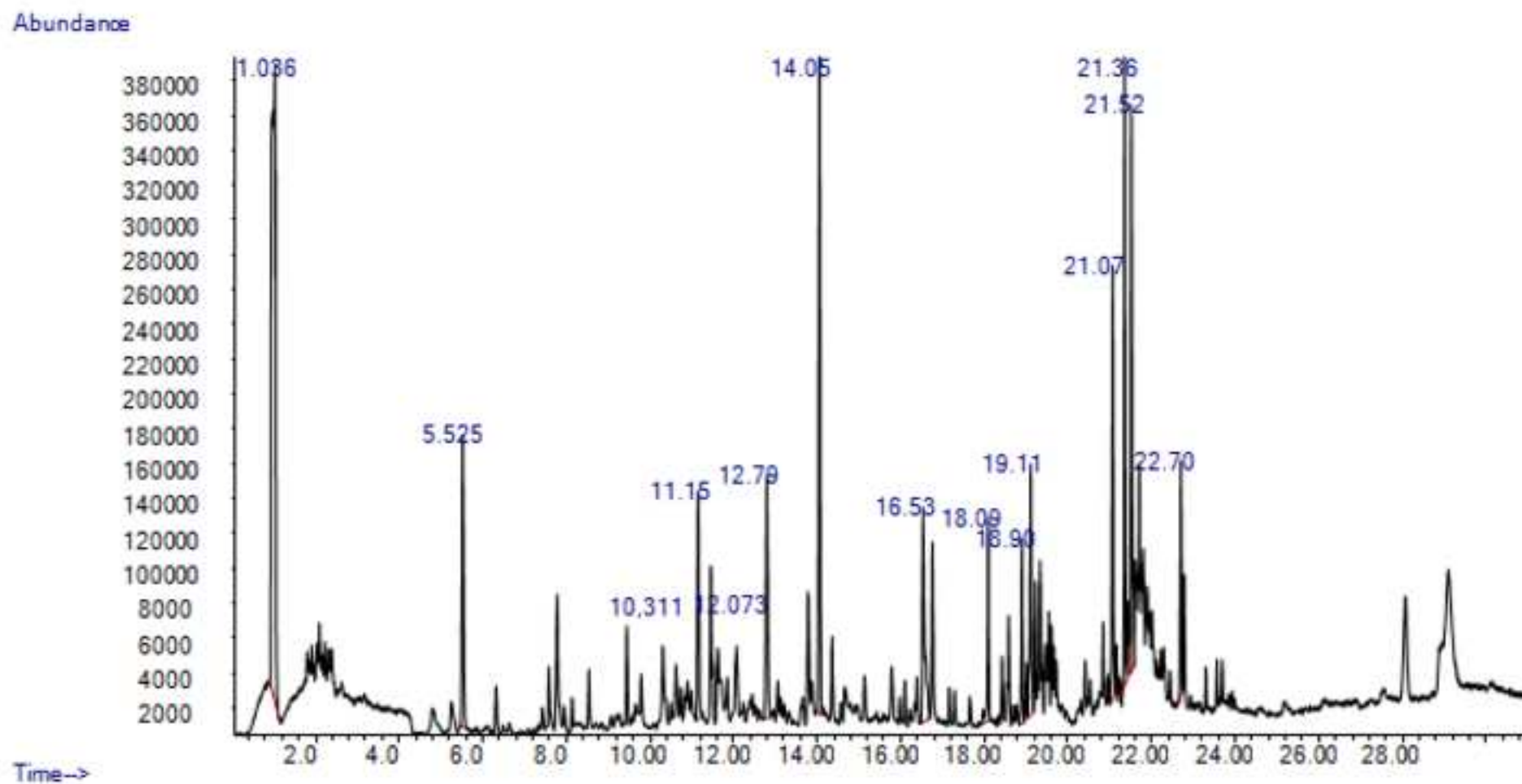
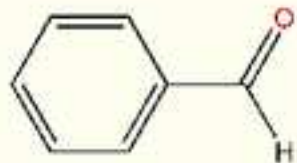


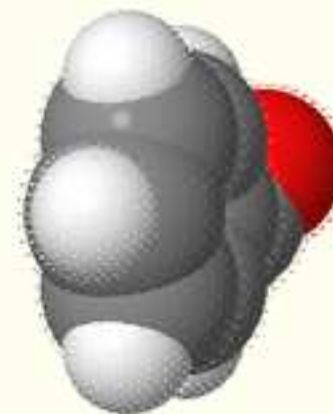
Figure 2

[Click here to access/download;Figure;Figure 2.png](#)


## Benzaldehyde

Formula: C<sub>7</sub>H<sub>6</sub>O  
CAS#: 100-52-7  
MW: 106.12

[MS] [NMR] [Kovats] [Synthesis] [Behavioural function] [Chemdraw]



Plant species emit Benzaldehyde

### Agaricales, Agaricaceae

*Agaricus smithii*

### Alismatales, Araceae

*Amorphophallus albispadix*  
*Amorphophallus albus*  
*Amorphophallus annulifer*  
*Amorphophallus titanum*  
*Amorphophallus zenkeri*  
*Anthurium antiquense*  
*Anthurium armeniacense*  
*Anthurium fragrantissimum*  
*Anthurium lindenbergii*  
*Anthurium ochranthum*  
*Anthurium roseospadix*  
*Arisaema ciliatum*  
*Arisaema erubescens*  
*Arisaema lobatum*  
*Arisaema speciosum*  
*Arisaema tortuosum*  
*Arum apulum*  
*Arum creticum*

### Asparagales, Orchidaceae

*Acacallis cyanea*  
*Acacallis superba*  
*Acineta superba*  
*Aerangis appendiculata*  
*Aerangis biloba*  
*Aerangis brachycarpa*  
*Aerangis confusa*  
*Aerangis distincta*  
*Aerangis fastuosa*  
*Aerangis kirkii*  
*Aerangis kotschyana*  
*Aerangis modesta*  
*Aerangis somaliensis*  
*Aeranthus grandiflora*  
*Aerides crassifolia*  
*Aerides fieldingii*  
*Aerides jackianum*  
*Aerides lawrenceae*  
*Aqanisia pulchella*  
*Anacamptis pyramidalis*

*Ancistrotichilus rothschildianus*  
*Angraecopsis amanienensis*  
*Angraecum aporoides*  
*Angraecum bossen*  
*Angraecum eburneum*  
*Angraecum eichlerianum*  
*Angraecum girvinae*  
*Angraecum sesquipedale*  
*Anguloa slowensis*  
*Anguloa uniflora*  
*Anguloa virginialis*  
*Ansellia gigantea*  
*Aspasia epidendroides*  
*Aspasia principissa*  
*Aspasia variegata*  
*Bifrenaria flagellaria*  
*Bifrenaria fuerstembergiana*  
*Bifrenaria thyrsanthina*  
*Bifrenaria wittii*  
*Bollea confertifolia*  
*Brassavola diabyana*

*Brassavola glauca*  
*Brassavola nodosa*  
*Brassavola sp.*  
*Brassavola tuberculata*  
*Brassia lobbia*  
*Brassia verucosa*  
*Catasetum discolor*  
*Catasetum expansum*  
*Catasetum fimbriatum*  
*Catasetum fuchsii*  
*Catasetum gnomus*  
*Catasetum integerrimum*  
*Catasetum longifolium*  
*Catasetum macroGLOSSUM*  
*Catasetum maculatum*  
*Catasetum russelliana*  
*Catasetum saccatum*  
*Catasetum tabulare*  
*Catasetum viridiflavum*  
*Cattleya araguaiensis*  
*Cattleya bicolor*  
*Cattleya deflexa*

<b>Name of Material/ Equipment</b>	<b>Company</b>	<b>Catalog Number</b>	<b>Comments/Description</b>
Bulkhead Union	Cole-Palmer	UX-06390-10	
FEP tubing	Cole-Palmer	UX-06407-60	
Gas Chromatography	Hewlett Packard	6890	
Glass Wool, Silanized	Sigma-Aldrich	20411	
Inlet liner	Agilent	5062-3587	
Mass Spectrometer	Hewlett Packard	5973	
Reynolds oven bag	Reynolds Consumer Products	Turkey size	
Tenax Porous Polymer Adsorbent	Sigma-Aldrich	11982	



1 Alewife Center #200  
Cambridge, MA 02140  
tel. 617.945.9051  
www.jove.com

## ARTICLE AND VIDEO LICENSE AGREEMENT

Title of Article:

Rapid Collection of Floral Fragrance volatiles using a headspace volatile collection technique for

Author(s):

Halegh A Ray, Charles J Stuhl, Jennifer L Gillett-Haufman  
GC-MS thermal desorption sample analysis

Item 1 (check one box): The Author elects to have the Materials be made available (as described at

<http://www.jove.com/author>) via: ☐ Standard Access ☒ Open Access

Item 2 (check one box):

☐

The Author is NOT a United States government employee.

☒

The Author is a United States government employee and the Materials were prepared in the course of his or her duties as a United States government employee.

☐

The Author is a United States government employee but the Materials were NOT prepared in the course of his or her duties as a United States government employee.

### ARTICLE AND VIDEO LICENSE AGREEMENT

1. **Defined Terms.** As used in this Article and Video License Agreement, the following terms shall have the following meanings: "**Agreement**" means this Article and Video License Agreement; "**Article**" means the article specified on the last page of this Agreement, including any associated materials such as texts, figures, tables, artwork, abstracts, or summaries contained therein; "**Author**" means the author who is a signatory to this Agreement; "**Collective Work**" means a work, such as a periodical issue, anthology or encyclopedia, in which the Materials in their entirety in unmodified form, along with a number of other contributions, constituting separate and independent works in themselves, are assembled into a collective whole; "**CRC License**" means the Creative Commons Attribution-Non Commercial-No Derivs 3.0 Unported Agreement, the terms and conditions of which can be found at: <http://creativecommons.org/licenses/by-nc-nd/3.0/legalcode>; "**Derivative Work**" means a work based upon the Materials or upon the Materials and other pre-existing works, such as a translation, musical arrangement, dramatization, fictionalization, motion picture version, sound recording, art reproduction, abridgment, condensation, or any other form in which the Materials may be recast, transformed, or adapted; "**Institution**" means the institution, listed on the last page of this Agreement, by which the Author was employed at the time of the creation of the Materials; "**JoVE**" means MyJoVE Corporation, a Massachusetts corporation and the publisher of *The Journal of Visualized Experiments*; "**Materials**" means the Article and / or the Video; "**Parties**" means the Author and JoVE; "**Video**" means any video(s) made by the Author, alone or in conjunction with any other parties, or by JoVE or its affiliates or agents, individually or in collaboration with the Author or any other parties, incorporating all or any portion of the Article, and in which the Author may or may not appear.

2. **Background.** The Author, who is the author of the Article, in order to ensure the dissemination and protection of the Article, desires to have the JoVE publish the Article and create and transmit videos based on the Article. In furtherance of such goals, the Parties desire to memorialize in this Agreement the respective rights of each Party in and to the Article and the Video.

3. **Grant of Rights in Article.** In consideration of JoVE agreeing to publish the Article, the Author hereby grants to JoVE, subject to **Sections 4 and 7** below, the exclusive, royalty-free, perpetual (for the full term of copyright in the Article, including any extensions thereto) license (a) to publish, reproduce, distribute, display and store the Article in all forms, formats and media whether now known or hereafter developed (including without limitation in print, digital and electronic form) throughout the world, (b) to translate the Article into other languages, create adaptations, summaries or extracts of the Article or other Derivative Works (including, without limitation, the Video) or Collective Works based on all or any portion of the Article and exercise all of the rights set forth in (a) above in such translations, adaptations, summaries, extracts, Derivative Works or Collective Works and (c) to license others to do any or all of the above. The foregoing rights may be exercised in all media and formats, whether now known or hereafter devised, and include the right to make such modifications as are technically necessary to exercise the rights in other media and formats. If the "Open Access" box has been checked in **Item 1** above, JoVE and the Author hereby grant to the public all such rights in the Article as provided in, but subject to all limitations and requirements set forth in, the CRC License.



## ARTICLE AND VIDEO LICENSE AGREEMENT

4. Retention of Rights in Article. Notwithstanding the exclusive license granted to JoVE in **Section 3** above, the Author shall, with respect to the Article, retain the non-exclusive right to use all or part of the Article for the non-commercial purpose of giving lectures, presentations or teaching classes, and to post a copy of the Article on the Institution's website or the Author's personal website, in each case provided that a link to the Article on the JoVE website is provided and notice of JoVE's copyright in the Article is included. All non-copyright intellectual property rights in and to the Article, such as patent rights, shall remain with the Author.

5. Grant of Rights in Video – Standard Access. This **Section 5** applies if the "Standard Access" box has been checked in **Item 1** above or if no box has been checked in **Item 1** above. In consideration of JoVE agreeing to produce, display or otherwise assist with the Video, the Author hereby acknowledges and agrees that, Subject to **Section 7** below, JoVE is and shall be the sole and exclusive owner of all rights of any nature, including, without limitation, all copyrights, in and to the Video. To the extent that, by law, the Author is deemed, now or at any time in the future, to have any rights of any nature in or to the Video, the Author hereby disclaims all such rights and transfers all such rights to JoVE.

6. Grant of Rights in Video – Open Access. This **Section 6** applies only if the "Open Access" box has been checked in **Item 1** above. In consideration of JoVE agreeing to produce, display or otherwise assist with the Video, the Author hereby grants to JoVE, subject to **Section 7** below, the exclusive, royalty-free, perpetual (for the full term of copyright in the Article, including any extensions thereto) license (a) to publish, reproduce, distribute, display and store the Video in all forms, formats and media whether now known or hereafter developed (including without limitation in print, digital and electronic form) throughout the world, (b) to translate the Video into other languages, create adaptations, summaries or extracts of the Video or other Derivative Works or Collective Works based on all or any portion of the Video and exercise all of the rights set forth in (a) above in such translations, adaptations, summaries, extracts, Derivative Works or Collective Works and (c) to license others to do any or all of the above. The foregoing rights may be exercised in all media and formats, whether now known or hereafter devised, and include the right to make such modifications as are technically necessary to exercise the rights in other media and formats. For any Video to which this Section 6 is applicable, JoVE and the Author hereby grant to the public all such rights in the Video as provided in, but subject to all limitations and requirements set forth in, the CRC License.

7. Government Employees. If the Author is a United States government employee and the Article was prepared in the course of his or her duties as a United States government employee, as indicated in **Item 2** above, and any of the licenses or grants granted by the Author hereunder exceed the scope of the 17 U.S.C. 403, then the rights granted hereunder shall be limited to the maximum rights permitted under such

statute. In such case, all provisions contained herein that are not in conflict with such statute shall remain in full force and effect, and all provisions contained herein that do so conflict shall be deemed to be amended so as to provide to JoVE the maximum rights permissible within such statute.

8. Likeness, Privacy, Personality. The Author hereby grants JoVE the right to use the Author's name, voice, likeness, picture, photograph, image, biography and performance in any way, commercial or otherwise, in connection with the Materials and the sale, promotion and distribution thereof. The Author hereby waives any and all rights he or she may have, relating to his or her appearance in the Video or otherwise relating to the Materials, under all applicable privacy, likeness, personality or similar laws.

9. Author Warranties. The Author represents and warrants that the Article is original, that it has not been published, that the copyright interest is owned by the Author (or, if more than one author is listed at the beginning of this Agreement, by such authors collectively) and has not been assigned, licensed, or otherwise transferred to any other party. The Author represents and warrants that the author(s) listed at the top of this Agreement are the only authors of the Materials. If more than one author is listed at the top of this Agreement and if any such author has not entered into a separate Article and Video License Agreement with JoVE relating to the Materials, the Author represents and warrants that the Author has been authorized by each of the other such authors to execute this Agreement on his or her behalf and to bind him or her with respect to the terms of this Agreement as if each of them had been a party hereto as an Author. The Author warrants that the use, reproduction, distribution, public or private performance or display, and/or modification of all or any portion of the Materials does not and will not violate, infringe and/or misappropriate the patent, trademark, intellectual property or other rights of any third party. The Author represents and warrants that it has and will continue to comply with all government, institutional and other regulations, including, without limitation all institutional, laboratory, hospital, ethical, human and animal treatment, privacy, and all other rules, regulations, laws, procedures or guidelines, applicable to the Materials, and that all research involving human and animal subjects has been approved by the Author's relevant institutional review board.

10. JoVE Discretion. If the Author requests the assistance of JoVE in producing the Video in the Author's facility, the Author shall ensure that the presence of JoVE employees, agents or independent contractors is in accordance with the relevant regulations of the Author's institution. If more than one author is listed at the beginning of this Agreement, JoVE may, in its sole discretion, elect not take any action with respect to the Article until such time as it has received complete, executed Article and Video License Agreements from each such author. JoVE reserves the right, in its absolute and sole discretion and without giving any reason therefore, to accept or decline any work submitted to JoVE. JoVE and its employees, agents and independent contractors shall have

## ARTICLE AND VIDEO LICENSE AGREEMENT

full, unfettered access to the facilities of the Author or of the Author's institution as necessary to make the Video, whether actually published or not. JoVE has sole discretion as to the method of making and publishing the Materials, including, without limitation, to all decisions regarding editing, lighting, filming, timing of publication, if any, length, quality, content and the like.

11. **Indemnification.** The Author agrees to indemnify JoVE and/or its successors and assigns from and against any and all claims, costs, and expenses, including attorney's fees, arising out of any breach of any warranty or other representations contained herein. The Author further agrees to indemnify and hold harmless JoVE from and against any and all claims, costs, and expenses, including attorney's fees, resulting from the breach by the Author of any representation or warranty contained herein or from allegations or instances of violation of intellectual property rights, damage to the Author's or the Author's institution's facilities, fraud, libel, defamation, research, equipment, experiments, property damage, personal injury, violations of institutional, laboratory, hospital, ethical, human and animal treatment, privacy or other rules, regulations, laws, procedures or guidelines, liabilities and other losses or damages related in any way to the submission of work to JoVE, making of videos by JoVE, or publication in JoVE or elsewhere by JoVE. The Author shall be responsible for, and shall hold JoVE harmless from, damages caused by lack of sterilization, lack of cleanliness or by contamination due to the making of a video by JoVE its employees, agents or independent contractors. All sterilization, cleanliness or decontamination procedures shall be solely the responsibility of the Author and shall be undertaken at the Author's

expense. All indemnifications provided herein shall include JoVE's attorney's fees and costs related to said losses or damages. Such indemnification and holding harmless shall include such losses or damages incurred by, or in connection with, acts or omissions of JoVE, its employees, agents or independent contractors.

12. **Fees.** To cover the cost incurred for publication, JoVE must receive payment before production and publication the Materials. Payment is due in 21 days of invoice. Should the Materials not be published due to an editorial or production decision, these funds will be returned to the Author. Withdrawal by the Author of any submitted Materials after final peer review approval will result in a US\$1,200 fee to cover pre-production expenses incurred by JoVE. If payment is not received by the completion of filming, production and publication of the Materials will be suspended until payment is received.

13. **Transfer, Governing Law.** This Agreement may be assigned by JoVE and shall inure to the benefits of any of JoVE's successors and assignees. This Agreement shall be governed and construed by the internal laws of the Commonwealth of Massachusetts without giving effect to any conflict of law provision thereunder. This Agreement may be executed in counterparts, each of which shall be deemed an original, but all of which together shall be deemed to me one and the same agreement. A signed copy of this Agreement delivered by facsimile, e-mail or other means of electronic transmission shall be deemed to have the same legal effect as delivery of an original signed copy of this Agreement.

A signed copy of this document must be sent with all new submissions. Only one Agreement required per submission.

### CORRESPONDING AUTHOR:

Name:

Haleigh Ray

Department:

Dept. of Entomology + Nematology

Institution:

University of Florida

Article Title:

Rapid collection of floral fragrance volatiles using a headspace volatile collection

Signature:

Haleigh Ray

Date:

8-6-18

technique for  
GC-MS chemical  
description  
sample analysis

Please submit a signed and dated copy of this license by one of the following three methods:

- 1) Upload a scanned copy of the document as a pdf on the JoVE submission site;
- 2) Fax the document to +1.866.381.2236;
- 3) Mail the document to JoVE / Attn: JoVE Editorial / 1 Alewife Center #200 / Cambridge, MA 02139

For questions, please email submissions@jove.com or call +1.617.945.9051





**College of Agricultural and Life Sciences**

Department of Entomology and Nematology

Bldg. 970, Natural  
Area Drive  
PO Box 110620  
Gainesville, FL  
32611-0620  
352-392-1901  
352-392-0190 Fax

JoVE Editorial Board,

We would like to submit the revisions for our paper, "Rapid collection of floral fragrance volatiles using a headspace volatile collection technique for GC-MS thermal desorption sample analysis". The comments from the reviewers and/or editor have been addressed, and the manuscript submitted shows the "Track Changes" for the revisions made. I have also uploaded a document titled "JoVE Final", which is a clean version of the document with the aforementioned changes accepted. In addition, we have added a figure (Figure 2), as well as a supplementary file, in order to better explain the results of these methods. Please let me know if there is anything else that I can do.

Thank you,

Haleigh A. Ray  
University of Florida  
Department of Entomology and Nematology  
hray12@ufl.edu

## Library Search Report

Data Path : D:\msdata\CS\Haleigh\  
Data File : UF1 8 AM 031615.D  
Acq On : 17 Mar 2015 8:03  
Operator :  
Sample : UF1 8 am 031615.D  
Misc :  
ALS Vial : 9 Sample Multiplier: 1

Search Libraries: C:\Database\NIST14.L Minimum Quality: 80

Unknown Spectrum: Apex  
Integration Events: RTE Integrator - LSCINT.P

Pk#	RT	Area%	Library/ID	Ref#	CAS#	Qual
1	0.967	5.52	C:\Database\NIST14.L			
			Propane	80	000074-98-6	3
			Propane	79	000074-98-6	3
			Propane	78	000074-98-6	3
2	1.801	0.21	C:\Database\NIST14.L			
			TATP	86153	017088-37-8	25
			Acetone	218	000067-64-1	9
			Acetone	214	000067-64-1	9
3	1.907	0.20	C:\Database\NIST14.L			
			Acetone	215	000067-64-1	10
			Acetone	214	000067-64-1	10
			Acetone	218	000067-64-1	10
4	2.036	0.37	C:\Database\NIST14.L			
			Acetone	218	000067-64-1	22
			Acetone	217	000067-64-1	12
			Acetone	214	000067-64-1	12
5	2.094	0.48	C:\Database\NIST14.L			
			TATP	86153	017088-37-8	56
			Acetone	218	000067-64-1	35
			Acetone	214	000067-64-1	25
6	2.153	0.25	C:\Database\NIST14.L			
			Acetone	218	000067-64-1	14
			Acetone	217	000067-64-1	10
			Acetone	215	000067-64-1	10
7	2.228	0.39	C:\Database\NIST14.L			
			TATP	86153	017088-37-8	43
			Acetone	217	000067-64-1	10
			Acetone	218	000067-64-1	10
8	2.388	0.28	C:\Database\NIST14.L			
			TATP	86153	017088-37-8	56
			O-Methylisourea	782	002440-60-0	39
			Acetone	218	000067-64-1	14



9	4.803	0.37	C:\Database\NIST14.L		
			Acetone	214	000067-64-1 43
			Acetic acid, sodium salt	1146	000127-09-3 38
			Methyl glyoxal	634	000078-98-8 37
10	5.246	0.22	C:\Database\NIST14.L		
			Silanol, trimethyl-	2269	001066-40-6 64
			Propanoic acid, TMS derivative	22476	016844-98-7 53
			Isopropyl alcohol, TMS derivative	14712	001825-64-5 45
11	5.524	3.68	C:\Database\NIST14.L		
			Ethyl Acetate	2042	000141-78-6 91
			Ethyl Acetate	2038	000141-78-6 90
			Ethyl Acetate	2040	000141-78-6 86
12	6.325	0.62	C:\Database\NIST14.L		
			Pentanal	1727	000110-62-3 80
			Pentanal	1728	000110-62-3 59
			Butanal, 3-methyl-	1765	000590-86-3 56
13	7.426	0.28	C:\Database\NIST14.L		
			1,3,5-Cycloheptatriene	2495	000544-25-2 42
			Toluene	2481	000108-88-3 38
			1,3,5-Cycloheptatriene	2496	000544-25-2 38
14	7.586	0.76	C:\Database\NIST14.L		
			2-Propanone, 1-hydroxy-	827	000116-09-6 45
			2-Propanone, 1-hydroxy-	825	000116-09-6 45
			2-Propanone, 1-hydroxy-	826	000116-09-6 9
15	7.784	1.92	C:\Database\NIST14.L		
			Hexanal	3832	000066-25-1 80
			Hexanal	3833	000066-25-1 76
			Hexanal	3830	000066-25-1 64
16	7.955	0.30	C:\Database\NIST14.L		
			Cyclopropane, pentyl-	6813	002511-91-3 87
			cis-1-Butyl-2-methylcyclopropane	6920	038851-69-3 81
			trans-1-Butyl-2-methylcyclopropane	6934	038851-70-6 81
17	8.136	0.37	C:\Database\NIST14.L		
			Heptane, 2,4-dimethyl-	12973	002213-23-2 50
			Sulfurous acid, dodecyl hexyl este	191786	1000309-13-4 50
			Octane	7760	000111-65-9 49
18	8.542	0.62	C:\Database\NIST14.L		
			Cyclotrisiloxane, hexamethyl-	85992	000541-05-9 90
			Cyclotrisiloxane, hexamethyl-	85994	000541-05-9 86
			Cyclotrisiloxane, hexamethyl-	85993	000541-05-9 64
19	9.450	1.07	C:\Database\NIST14.L		
			Heptanal	7559	000111-71-7 96
			Heptanal	7563	000111-71-7 94
			Heptanal	7562	000111-71-7 90
20	9.792	0.49	C:\Database\NIST14.L		
			Nonane	12937	000111-84-2 76
			Nonane	12939	000111-84-2 76

		Nonane	12938 000111-84-2 64
21	10.311	1.15 C:\Database\NIST14.L Benzaldehyde Benzaldehyde Benzaldehyde	5152 000100-52-7 94 5148 000100-52-7 94 5151 000100-52-7 93
22	10.498	0.24 C:\Database\NIST14.L Benzene, propyl- Benzene, propyl- Benzene, propyl-	9586 000103-65-1 68 9585 000103-65-1 58 9587 000103-65-1 52
23	10.620	0.71 C:\Database\NIST14.L Benzene, 1-ethyl-2-methyl- Benzene, 1-ethyl-2-methyl- Benzene, 1-ethyl-3-methyl-	9606 000611-14-3 93 9603 000611-14-3 93 9604 000620-14-4 87
24	10.743	0.37 C:\Database\NIST14.L Mesitylene 2,4-Nonadiyne Benzene, 1,2,3-trimethyl-	9581 000108-67-8 76 9583 063621-15-8 76 9599 000526-73-8 76
25	10.839	0.25 C:\Database\NIST14.L Nonane, 4-methyl- Nonane, 4-methyl- Nonane, 4-methyl-	19664 017301-94-9 87 19658 017301-94-9 81 19665 017301-94-9 58
26	10.898	0.43 C:\Database\NIST14.L Hexadecane Decane, 6-ethyl-2-methyl- 2,2-Dimethyl-3-octanone	89841 000544-76-3 43 51469 062108-21-8 43 29209 005340-64-7 43
27	10.989	0.23 C:\Database\NIST14.L Phenol 3-Methylpyridazine 2-Vinylfuran	2619 000108-95-2 53 2605 001632-76-4 49 2622 001487-18-9 49
28	11.155	2.87 C:\Database\NIST14.L Mesitylene Benzene, 1,2,3-trimethyl- Benzene, 1-ethyl-3-methyl-	9578 000108-67-8 74 9592 000526-73-8 74 9609 000620-14-4 70
29	11.459	1.70 C:\Database\NIST14.L Decane Decane Decane	19648 000124-18-5 96 19649 000124-18-5 95 19651 000124-18-5 94
30	11.513	0.46 C:\Database\NIST14.L Cyclotetrasiloxane, octamethyl- Cyclotetrasiloxane, octamethyl- 2,5-Dihydroxyacetophenone, 2TMS de rivative	155949 000556-67-2 90 155952 000556-67-2 86 155021 1000352-83-7 59
31	11.619	1.22 C:\Database\NIST14.L Benzeneacetaldehyde Benzeneacetaldehyde Benzeneacetaldehyde	9549 000122-78-1 55 9551 000122-78-1 55 9550 000122-78-1 55

32	11.849	0.54	C:\Database\NIST14.L		
			D-Limonene	16046	005989-27-5 50
			1-Cyclohexyl-2-buten-1-ol (c,t)	27614	079605-62-2 46
			Limonene	16027	000138-86-3 46
33	12.073	1.21	C:\Database\NIST14.L		
			.beta.-Ocimene	16062	013877-91-3 60
			1,3,6-Octatriene, 3,7-dimethyl-, (Z)-	16175	003338-55-4 60
			3-Carene	16037	013466-78-9 53
34	12.447	0.19	C:\Database\NIST14.L		
			Dodecane, 2,6,11-trimethyl-	76619	031295-56-4 90
			Dodecane, 2,6,11-trimethyl-	76621	031295-56-4 81
			Decane, 3,7-dimethyl-	39995	017312-54-8 81
35	12.795	3.19	C:\Database\NIST14.L		
			Nonanal	20435	000124-19-6 86
			Nonanal	20436	000124-19-6 83
			Nonanal	20437	000124-19-6 80
36	13.062	0.47	C:\Database\NIST14.L		
			Undecane	29356	001120-21-4 97
			Undecane	29357	001120-21-4 93
			Undecane	29354	001120-21-4 87
37	13.126	0.19	C:\Database\NIST14.L		
			Decane, 3,6-dimethyl-	40000	017312-53-7 80
			Hexadecane, 7,9-dimethyl-	115564	021164-95-4 72
			Octane, 5-ethyl-2-methyl-	29382	062016-18-6 68
38	13.660	0.32	C:\Database\NIST14.L		
			Butylphosphonic acid, di(4-methoxy	194995	1000323-49-3 53
			-2-methylbutyl) ester		
			Thiocyanic acid, propyl ester	4121	004251-16-5 53
			Succinic acid, octyl propyl ester	132412	1000324-95-5 47
39	13.783	1.87	C:\Database\NIST14.L		
			Benzeneacetic acid, methyl ester	25201	000101-41-7 91
			Benzeneacetic acid, methyl ester	25199	000101-41-7 91
			Benzeneacetic acid, methyl ester	25197	000101-41-7 91
40	13.879	0.35	C:\Database\NIST14.L		
			1-(1-Methoxypropan-2-yloxy)propan-	56551	1000367-08-6 64
			2-yl acetate		
			Silane, trimethyl-	808	000993-07-7 59
			1-(1-Methoxypropan-2-yloxy)propan-	120926	1000367-13-1 59
			2-yl heptanoate		
41	14.055	6.75	C:\Database\NIST14.L		
			Cyclopentasiloxane, decamethyl-	220036	000541-02-6 91
			N-(Trifluoroacetyl)-O,O',O''-tris(t	265635	054135-51-2 38
			rimethylsilyl)epinephrine		
			N-(Trifluoroacetyl)-O,O',O''-tris(	263373	1000072-26-3 38
			trimethylsilyl)norepinephrine		
42	14.360	1.00	C:\Database\NIST14.L		
			Decanal	29131	000112-31-2 86

			Decanal	29132	000112-31-2	80
			13-Methyltetradecanal	89788	075853-51-9	52
43	14.574	0.26	C:\Database\NIST14.L			
			Octane	7759	000111-65-9	80
			Carbonic acid, decyl vinyl ester	91176	1000383-25-7	64
			Hexane, 3,3-dimethyl-	7774	000563-16-6	59
44	14.664	0.70	C:\Database\NIST14.L			
			Caprolactam	7051	000105-60-2	90
			Caprolactam	7052	000105-60-2	87
			Caprolactam	7050	000105-60-2	87
45	15.129	0.51	C:\Database\NIST14.L			
			1-Dodecene	38260	000112-41-4	58
			1-Pentadecene	74570	013360-61-7	53
			2-Decenal, (E)-	27514	003913-81-3	52
46	15.786	0.77	C:\Database\NIST14.L			
			Octane, 2-methyl-	12947	003221-61-2	38
			Pentadecane, 2,6,10,14-tetramethyl	128855	001921-70-6	38
			Pentadecane, 2,6,10,14-tetramethyl	128857	001921-70-6	35
47	15.979	0.26	C:\Database\NIST14.L			
			Tridecane	51395	000629-50-5	87
			Tridecane	51393	000629-50-5	87
			Hexadecane	89842	000544-76-3	86
48	16.101	0.45	C:\Database\NIST14.L			
			1-Undecene, 7-methyl-	38300	074630-42-5	64
			2-Methyl-Z-4-tetradecene	74578	1000130-78-3	52
			Hexadecanoic acid, 2-hydroxy-, methyl ester	146257	016742-51-1	52
49	16.219	0.18	C:\Database\NIST14.L			
			Octane, 4-methyl-	12945	002216-34-4	52
			Diglycolic acid, isohexyl 2-methyl butyl ester	147767	1000381-81-5	47
			Cyclohexane, 1,1-dimethyl-2-propyl	27977	081983-71-3	46
50	16.336	0.21	C:\Database\NIST14.L			
			Cyclopentane, hexyl-	27934	004457-00-5	49
			Cyclohexane, 1,2,4-trimethyl-	11829	002234-75-5	45
			1-Pentadecene	74570	013360-61-7	38
51	16.385	0.44	C:\Database\NIST14.L			
			Nonane, 1-iodo-	115703	004282-42-2	72
			Hexadecane, 1-iodo-	206779	000544-77-4	64
			Tetradecane, 1-iodo-	181794	019218-94-1	64
52	16.534	2.89	C:\Database\NIST14.L			
			2-Butyramido-2-[4-chloro-2-(trifluoromethyl)anilino]-1,1,1,3,3,3-hexafluoropropane	250555	1000224-20-4	32
			Cyclohexasiloxane, dodecamethyl-	254923	000540-97-6	25
			Cyclohexasiloxane, dodecamethyl-	254924	000540-97-6	22
53	16.759	1.96	C:\Database\NIST14.L			

			Butanoic acid, butyl ester	21449	000109-21-7	74
			Butanoic acid, hexyl ester	41263	002639-63-6	64
			Propanoic acid, 2-methyl-, 3-hydroxy-2,2,4-trimethylpentyl ester	79620	000077-68-9	64
54	17.159	0.40	C:\Database\NIST14.L			
			Tridecanal	63515	010486-19-8	76
			Undecanal	39874	000112-44-7	74
			Bacchotricuneatin c	199015	066563-30-2	58
55	17.293	0.30	C:\Database\NIST14.L			
			Pentadecane	76609	000629-62-9	46
			Sulfurous acid, butyl decyl ester	137914	1000309-17-7	43
			Sulfurous acid, butyl undecyl este	151433	1000309-17-8	43
56	17.651	0.32	C:\Database\NIST14.L			
			Hexasiloxane, tetradecamethyl-	258608	000107-52-8	50
			Trisiloxane, 1,1,1,5,5,5-hexamethy	229156	003555-47-3	43
			1-3,3-bis[(trimethylsilyl)oxy]-			
			Benzyloxyamine, 2TMS derivative	127249	079208-41-6	38
57	18.094	1.69	C:\Database\NIST14.L			
			4-(Adamantan-1-ylamino)-2,5,6-tric	209305	339060-92-3	50
			hloro-nicotinonitrile			
			1-Adamantanecarboxamide, N,N-dimet	71591	001502-00-7	40
			hyl-,			
			1,2-Benzisothiazole-3-propanoic ac	71226	050565-45-2	40
			id			
58	18.425	0.85	C:\Database\NIST14.L			
			2,4-Di-tert-butylphenol	70634	000096-76-4	96
			2,4-Di-tert-butylphenol	70632	000096-76-4	95
			Phenol, 2,5-bis(1,1-dimethylethyl)	70651	005875-45-6	87
59	18.537	0.43	C:\Database\NIST14.L			
			Eicosane	142239	000112-95-8	87
			Hexadecane	89842	000544-76-3	87
			Tetracosane	195673	000646-31-1	86
60	18.580	1.07	C:\Database\NIST14.L			
			Heptadecane	102600	000629-78-7	72
			Tridecane, 1-iodo-	168440	035599-77-0	64
			Undecane, 2-methyl-	39986	007045-71-8	64
61	18.692	0.14	C:\Database\NIST14.L			
			Octadecane	115546	000593-45-3	83
			Nonadecane	128834	000629-92-5	72
			Tetradecane, 1-iodo-	181793	019218-94-1	72
62	18.773	0.15	C:\Database\NIST14.L			
			6,6-Diethylhoctadecane	169414	1000360-41-8	80
			Tricosane, 2-methyl-	195680	001928-30-9	74
			Eicosane	142240	000112-95-8	74
63	18.901	1.77	C:\Database\NIST14.L			
			Cyclohexane, 1,2,4-trimethyl-	11827	002234-75-5	45
			Cyclohexane, 1,2,4-trimethyl-	11829	002234-75-5	43
			Cyclopentane, (2-methylbutyl)-	18472	053366-38-4	38

64	19.008	0.48	C:\Database\NIST14.L Cyclohexane, 1,2,4-trimethyl- Decane, 3,7-dimethyl- Oxalic acid, 6-ethyloct-3-yl heptyl ester	11829 002234-75-5 30 39995 017312-54-8 30 186278 1000309-34-5 27
65	19.109	2.99	C:\Database\NIST14.L Octadecane, 1-iodo- Octane, 2-methyl- Heptacosane	226953 000629-93-6 80 12950 003221-61-2 68 227469 000593-49-7 64
66	19.227	1.24	C:\Database\NIST14.L Octane, 5-ethyl-2-methyl- Tetracosane, 1-iodo- Heptadecane	29382 062016-18-6 76 260060 1000406-32-0 72 102599 000629-78-7 64
67	19.285	0.15	C:\Database\NIST14.L Tetratetracontane Triacosane, 1-iodo- 2-methyloctacosane	273584 007098-22-8 80 270745 1000406-32-3 80 242320 1000376-72-8 72
68	19.333	1.27	C:\Database\NIST14.L Hexadecane Heptacosane Eicosane, 1-iodo-	89844 000544-76-3 86 227469 000593-49-7 86 241993 1000406-31-8 83
69	19.381	0.34	C:\Database\NIST14.L Hexacosane, 1-iodo- Hexacosane Docosane, 1-iodo-	265262 1000406-32-1 72 217890 000630-01-3 72 252495 1000406-31-9 64
70	19.462	0.58	C:\Database\NIST14.L Tetracosane Tetracosane Hexadecane	195673 000646-31-1 80 195672 000646-31-1 72 89844 000544-76-3 72
71	19.542	0.85	C:\Database\NIST14.L Sulfurous acid, hexadecyl 2-pentyl ester Sulfurous acid, dodecyl 2-pentyl ester 2,2,4-Trimethyl-1,3-pentanediol diisobutyrate	224830 1000309-16-5 64 178523 1000309-16-1 64 146067 006846-50-0 58
72	19.606	0.68	C:\Database\NIST14.L Heneicosane Docosane Octadecane, 1-iodo-	155888 000629-94-7 80 169409 000629-97-0 72 226952 000629-93-6 59
73	19.665	0.49	C:\Database\NIST14.L 2H-Pyran, tetrahydro-2-(12-pentadecyloxy)- Octadecane Heptadecane	167395 056666-38-7 92 115545 000593-45-3 91 102600 000629-78-7 87
74	19.718	0.41	C:\Database\NIST14.L Hexadecane	89843 000544-76-3 94

			Hexadecane	89840	000544-76-3	94
			Hexadecane	89841	000544-76-3	94
75	20.407	0.53	C:\Database\NIST14.L (4aS,8aS)-8-Isopentyl-4,4,7,8a-tetramethyl-1,2,3,4,4a,5,6,8a-octahydronaphthalene	123189	220766-79-0	38
			Silane, 1,3-decadiynyltrimethyl-	70601	084751-17-7	38
			Oxirane, [[4-(1,1-dimethylethyl)phenoxy]methyl]-	70543	003101-60-8	35
76	20.530	0.20	C:\Database\NIST14.L Nonahexacontanoic acid	276185	040710-32-5	49
			Dodecane, 3-methyl-	51413	017312-57-1	43
			Sulfurous acid, butyl tetradecyl ester	191788	1000309-18-1	78
77	20.845	0.58	C:\Database\NIST14.L Heptadecane	102599	000629-78-7	94
			Tridecane, 1-iodo-	168440	035599-77-0	93
			Hexadecane	89842	000544-76-3	90
78	20.963	0.27	C:\Database\NIST14.L Hentriacontane	252712	000630-04-6	87
			Pentadecane, 2,6,10,14-tetramethyl	128856	001921-70-6	86
			Tetracosane	195673	000646-31-1	86
79	21.070	4.36	C:\Database\NIST14.L Heneicosane	155888	000629-94-7	90
			Heptacosane	227469	000593-49-7	90
			Octacosane	235614	000630-02-4	86
80	21.166	0.44	C:\Database\NIST14.L 1-Iodo-2-methylundecane	154846	073105-67-6	90
			Docosane, 1-iodo-	252495	1000406-31-9	87
			2-Bromotetradecane	136205	074036-95-6	86
81	21.219	0.20	C:\Database\NIST14.L Heptadecane	102599	000629-78-7	91
			Octadecane, 1-iodo-	226952	000629-93-6	86
			Hexacosane, 1-iodo-	265262	1000406-32-1	86
82	21.358	6.04	C:\Database\NIST14.L Triacontyl pentafluoropropionate	272551	1000351-80-0	52
			Octacosyl heptafluorobutyrate	273239	1000351-83-6	52
			Hexacosyl pentafluoropropionate	269275	1000351-81-2	50
83	21.454	0.72	C:\Database\NIST14.L Cyclohexane, 1-ethyl-2-propyl-	27962	062238-33-9	41
			Pentadecane	76609	000629-62-9	38
			Hexadecane	89842	000544-76-3	30
84	21.518	5.49	C:\Database\NIST14.L Heneicosane	155888	000629-94-7	72
			2-Bromo dodecane	109279	013187-99-0	64
			Hexane, 2,3,4-trimethyl-	12990	000921-47-1	52
85	21.615	0.89	C:\Database\NIST14.L			

			2-Bromotetradecane	136205	074036-95-6	86
			Decane, 3,8-dimethyl-	40006	017312-55-9	80
			2-methyloctacosane	242320	1000376-72-8	78
86	21.711	1.11	C:\Database\NIST14.L			
			Heptacosane	227469	000593-49-7	91
			2-Bromo dodecane	109279	013187-99-0	90
			Hexadecane	89838	000544-76-3	90
87	21.764	0.43	C:\Database\NIST14.L			
			Hexadecyl octyl ether	209173	1000406-38-6	64
			11-Methyltricosane	195679	027538-41-6	64
			Carbonic acid, decyl nonyl ester	186528	1000383-15-8	64
88	21.807	0.90	C:\Database\NIST14.L			
			Hexadecane, 2,6,11,15-tetramethyl-	142256	000504-44-9	90
			2-Bromotetradecane	136205	074036-95-6	80
			Hexacosane	217890	000630-01-3	80
89	21.898	0.82	C:\Database\NIST14.L			
			Hentriacontane	252712	000630-04-6	80
			Tetracosane	195673	000646-31-1	74
			Dotriacontane, 1-iodo-	272203	1000406-32-4	72
90	21.951	0.24	C:\Database\NIST14.L			
			Heptadecane	102599	000629-78-7	94
			Pentacosane	207499	000629-99-2	91
			Eicosane	142238	000112-95-8	91
91	22.010	0.22	C:\Database\NIST14.L			
			Tetratriacontane, 17-hexadecyl-	275157	055256-07-0	58
			Carbonic acid, octadecyl vinyl ester	197481	1000382-54-4	58
			Hexacosane	217890	000630-01-3	49
92	22.229	0.20	C:\Database\NIST14.L			
			Triacontane, 1-bromo-	266454	004209-22-7	72
			Tetratetracontane	273584	007098-22-8	72
			Decane, 3,8-dimethyl-	40006	017312-55-9	70
93	22.293	0.78	C:\Database\NIST14.L			
			Ethanone, 1-(5,6,7,8-tetrahydro-2,8,8-trimethyl-4H-cyclohepta[b]furan-5-yl)-	83411	071596-88-8	52
			2,5-di-tert-Butyl-1,4-benzoquinone	83301	002460-77-7	42
			3-Acetylphenanthrene	83713	002039-76-1	25
94	22.448	0.32	C:\Database\NIST14.L			
			Homosalate	122907	000118-56-9	93
			Homosalate	122906	000118-56-9	81
			3,3,5-Trimethylcyclohexyl acrylate	61569	087954-40-3	22
95	22.704	2.16	C:\Database\NIST14.L			
			Homosalate	122906	000118-56-9	99
			Homosalate	122907	000118-56-9	98
			Homosalate	122908	000118-56-9	76
96	22.784	1.24	C:\Database\NIST14.L			



7,9-Di-tert-butyl-1-oxaspiro(4,5)d 136515 082304-66-3 95  
 eca-6,9-diene-2,8-dione  
 7,9-Di-tert-butyl-1-oxaspiro(4,5)d 136513 082304-66-3 93  
 eca-6,9-diene-2,8-dione  
 Ethanone, 1-(5,6,7,8-tetrahydro-2, 83411 071596-88-8 56  
 8,8-trimethyl-4H-cyclohepta[b]fura  
 n-5-yl)-

97 23.297 0.35 C:\Database\NIST14.L  
 Hentriacontane 252711 000630-04-6 90  
 Octacosane 235614 000630-02-4 87  
 Docosane 169409 000629-97-0 86

98 23.564 0.36 C:\Database\NIST14.L  
 Sulfurous acid, butyl octadecyl es 233252 1000309-18-5 46  
 ter  
 Cyclohexane, 1-ethyl-2-propyl- 27962 062238-33-9 38  
 Octadecane, 1-(ethenyloxy)- 155862 000930-02-9 30

99 23.693 0.36 C:\Database\NIST14.L  
 2-Methylhexacosane 227470 001561-02-0 90  
 Docosane, 1-iodo- 252495 1000406-31-9 87  
 Eicosane, 1-iodo- 241993 1000406-31-8 86

100 28.068 1.45 C:\Database\NIST14.L  
 Eicosane 142239 000112-95-8 86  
 Eicosane, 1-iodo- 241993 1000406-31-8 81  
 Hentriacontane 252712 000630-04-6 81

RGB-dataanalysis.M Wed Aug 01 09:07:03 2018




---




---




---




---




---




---

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION 

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION 

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION



QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION



QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION



QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION QUESTION

QUESTION 



\_\_\_\_\_



\_\_\_\_\_



\_\_\_\_\_



\_\_\_\_\_



\_\_\_\_\_



\_\_\_\_\_



\_\_\_\_\_



\_\_\_\_\_



\_\_\_\_\_



\_\_\_\_\_